

Kaltgasspritzen: Prinzip und Einsatzpotentiale für hitzeempfindliche Werkstoffe

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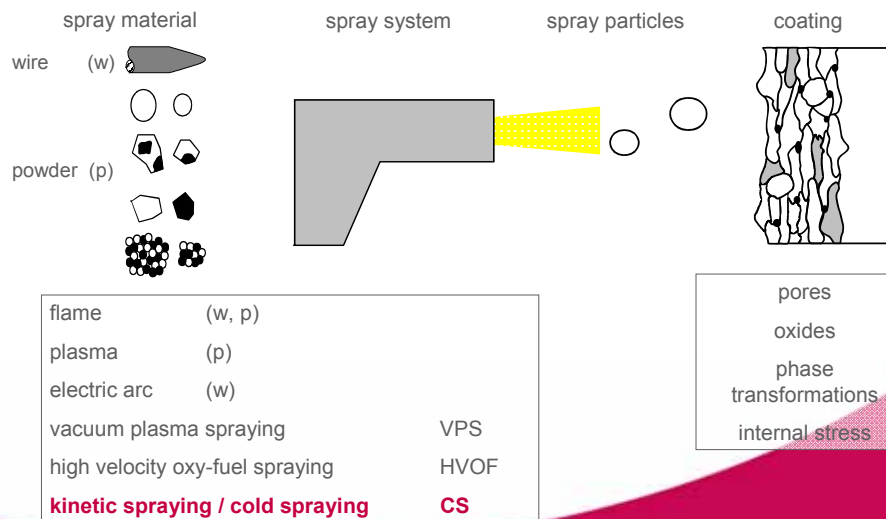
Workshop: Metallische Massivgläser, Hanau, Germany, April 28th, 2011

Overview of Coating Techniques

Typical Coating Thickness [µm]	Technique	Deposition from	Substrate Temperature [°C]	Typical Material
0,1 - 2,0	CVD / PVD	gas ●	CVD (300 - 1000) ● PVD (100 - 400) ●	TiC,....
0,5 - 150	electro / electroless plating	electrolyt ●	30 - 90 ●	Ni, Cr, Cu, metallic alloys
100 - 500	thermal spraying	molten alloys ●	100 - 250 ●	metallic alloys, ceramics, composites
50 - 1000 (max. 50.000)	cold spraying	solid alloy ●	30 - 200 ●	ductile materials, composites
2000 - 10.000	weld cladding	molten alloys ●	> 1000 ●	alloyed steels, hard alloys, (composites)
500 - 50.000	cold rolling, explosive cladding	solid alloys ●	30 - 150 ●	ductile materials, (Al - alloys, steels)

hot: ● „warm“: ● cold: ●

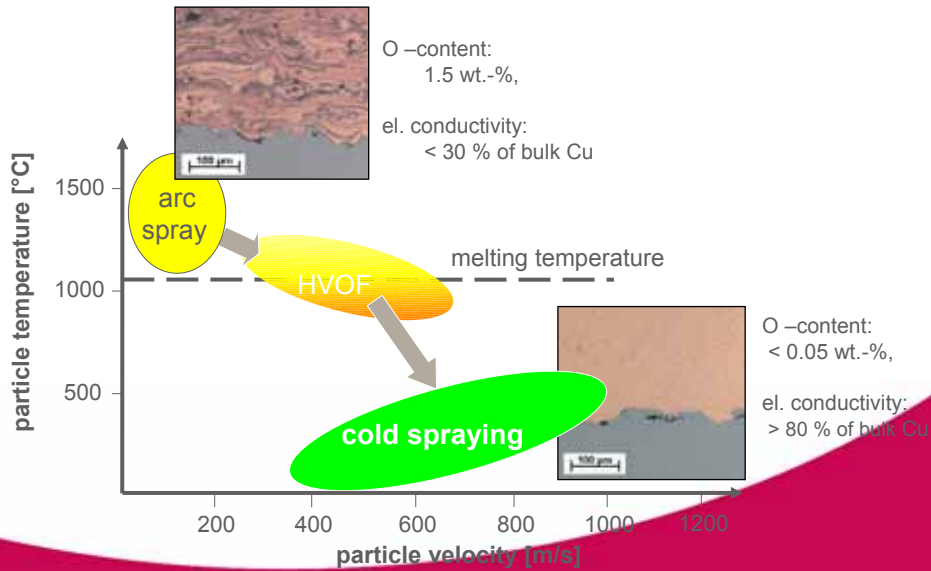
The Thermal Spray Process



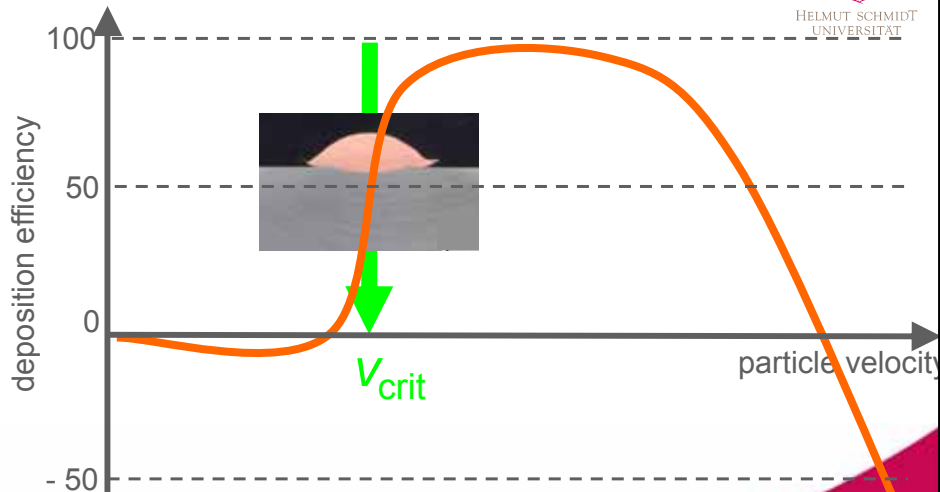
Heat Sensitive Materials

- metastable phases (amorphous, quasi-crystalline)
 - non equilibrium microstructures (nano-crystalline)
 - highly oxidative alloys
 - easy to process as powder
 - difficult to densify to massive parts
- possible solution:
- replacing thermal by kinetic energy input (like in explosive compaction)

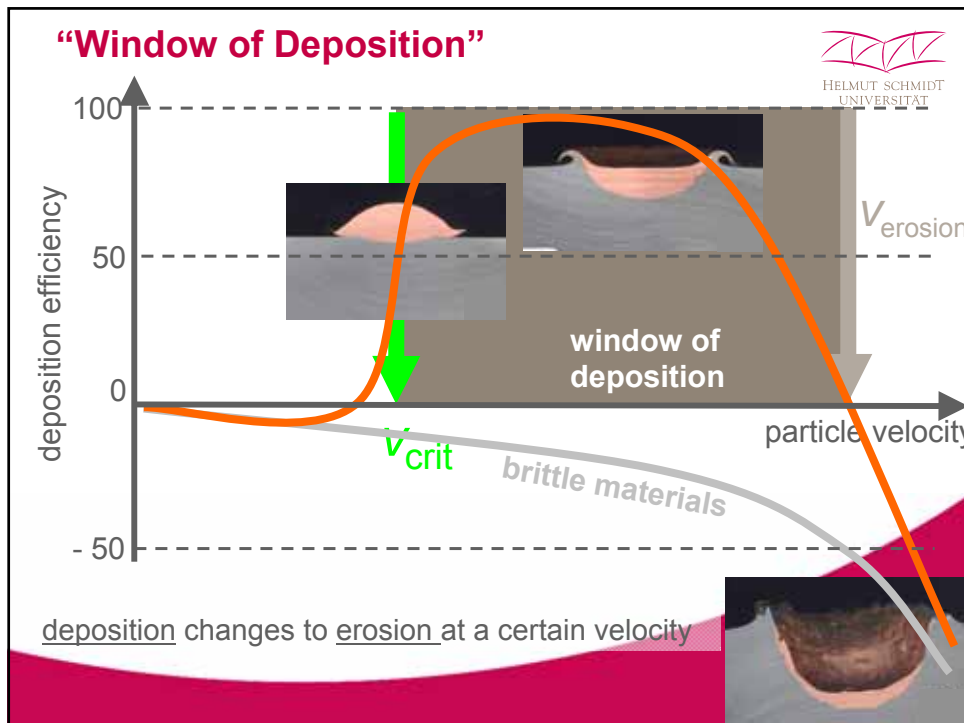
Influence on Coating Properties: Cu



Impact Effects: Deposition and Erosion



at a certain velocity particles start to stick
 ✓ at a DE of 50 % the critical velocity is defined



Requirements: Powder

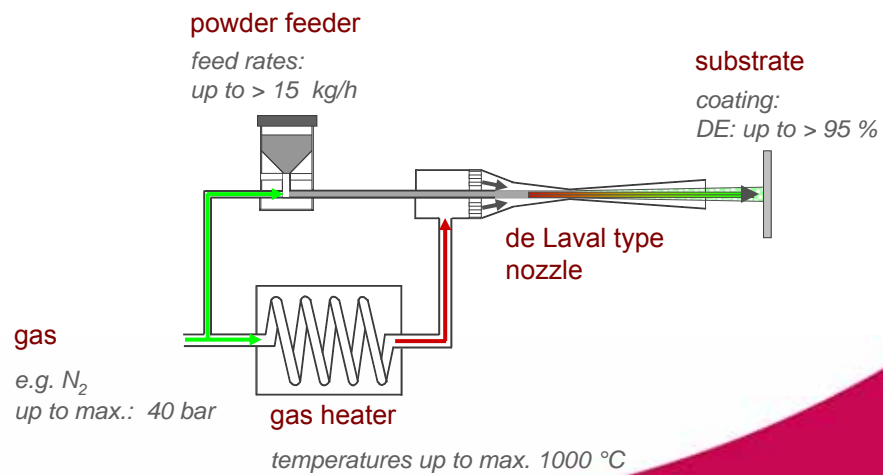
sizes: typically 15 to 50 μm , very low oxygen content

amorphous ZrCuAlNi alloy,
sizes < 45 μm

amorphous FeCrCoMoBC alloy,
sizes < 45 μm

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Requirements: Energy supply by Cold / Kinetic Spraying



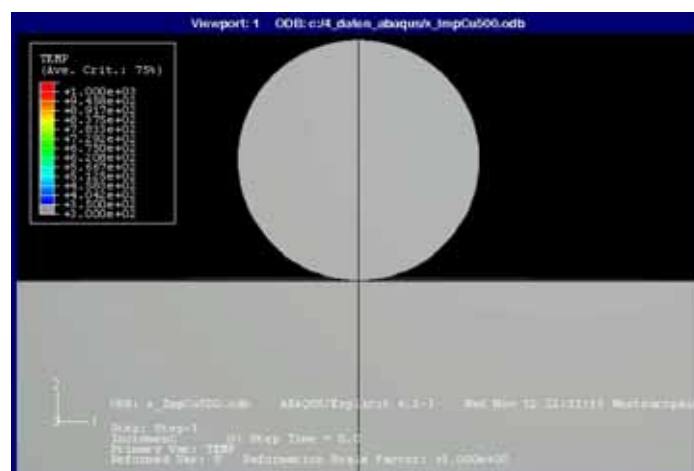
General Features: Cu as an Example

Example: Cold Spraying of Copper

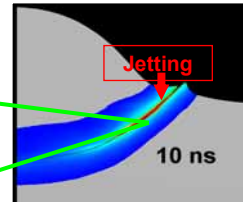
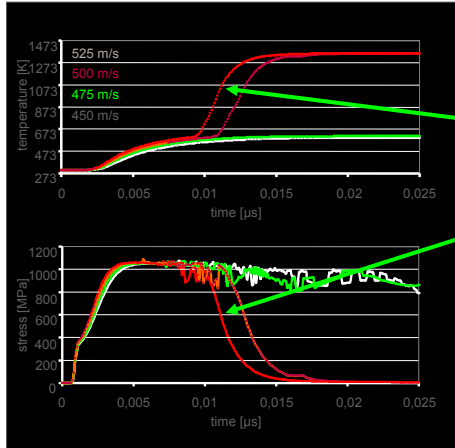


Modeling: Particle Impact Cu/Cu

temperature field, adiabatic calculation, $v = 500$ m/s



Bonding Mechanisms / Critical Velocities: Bonding by shear instabilities



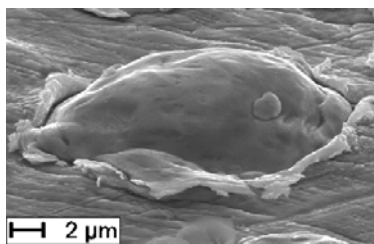
thermal softening and
break-down of local
stress at interfaces
by viscous like flow

results from modeling
impacts of Cu on Cu

time scales:

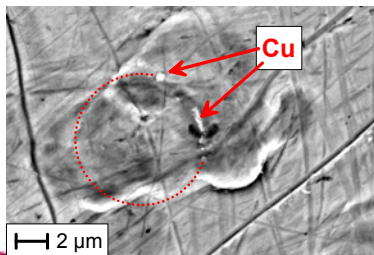
heating / quenching: $\sim 10^9$ /s
deformation: $\sim 10^9$ /s

Particle Impact - Exp. Observations



copper on copper

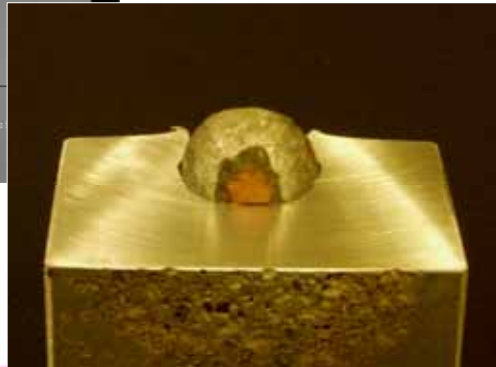
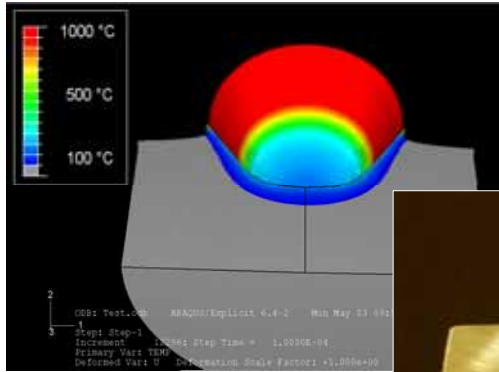
jet formation indicates
localized plastic flow of
material at the interface.



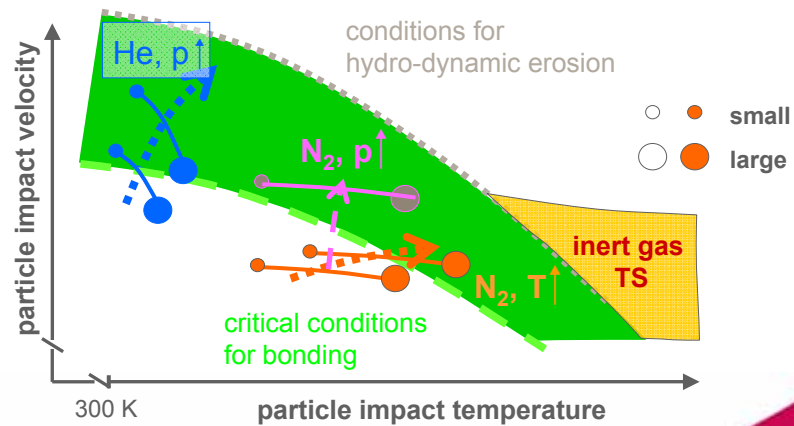
copper on steel

bonding extends from the rim of
crater inwards – never reaches
centre.

Area of Bonding



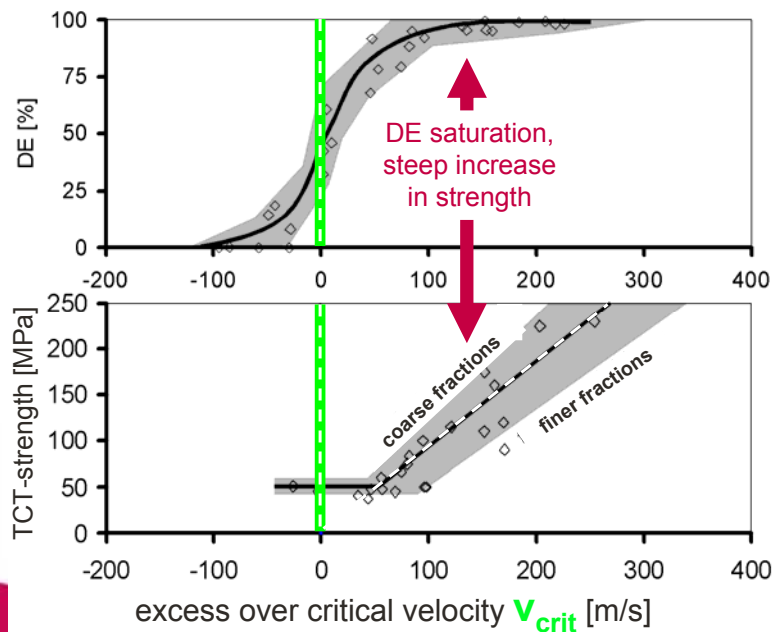
Process Improvements: Window of Sprayability



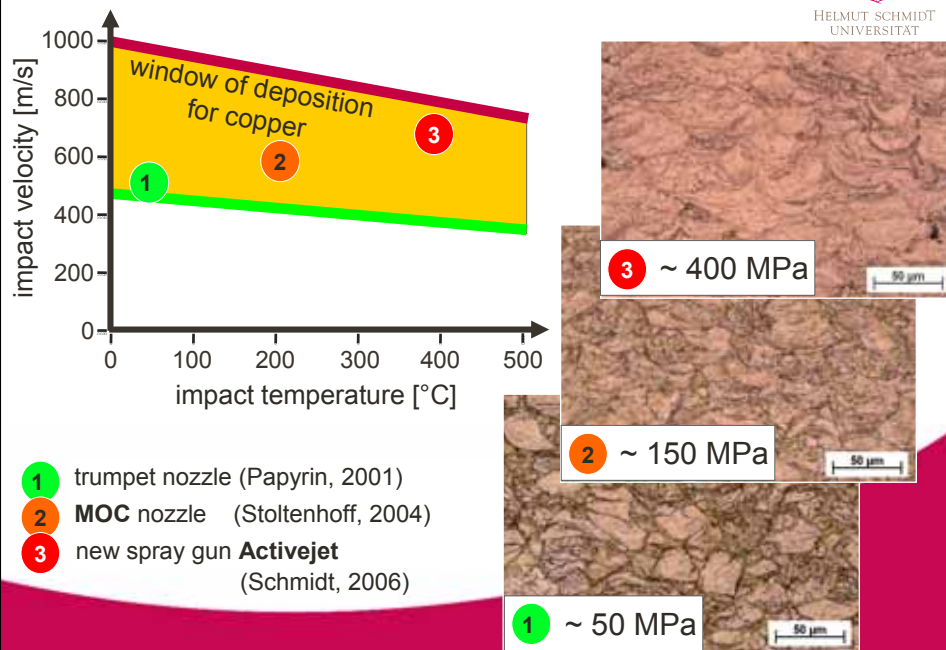
He route: very fast, cold impact condition (costly)

N₂ route: fast, warm impact condition (reactions?)

DE & UTS = function of impact conditions

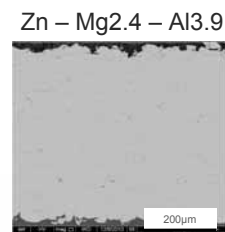
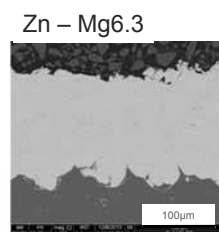
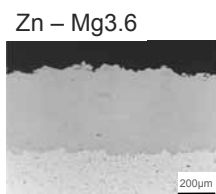
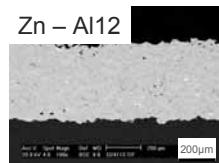
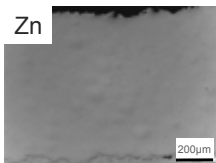


Impact conditions \Rightarrow microstructure & properties

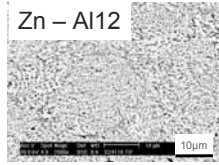
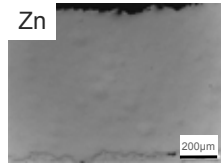


Zn-Coatings for Laser Gravure

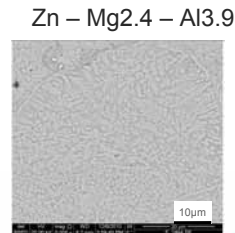
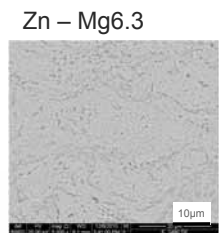
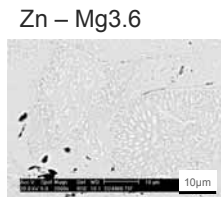
Zn-, Zn-Alloy- Coating Microstructures



Zn-, Zn-Alloy- Coating Microstructures

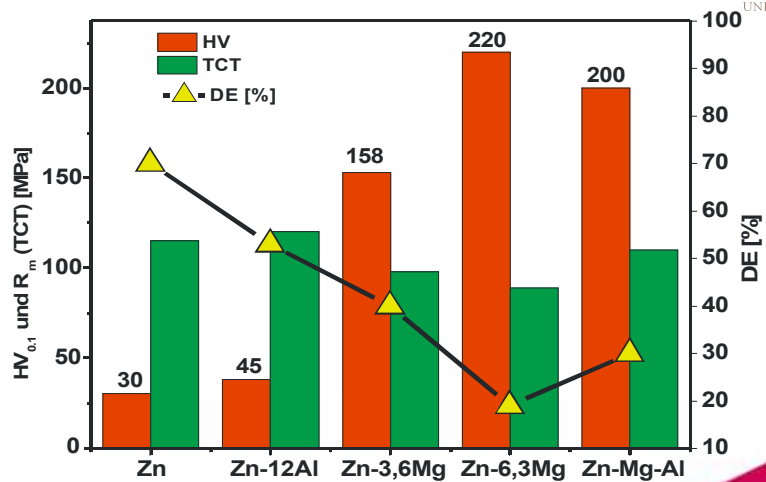


Porosity 0 – 0.3%



- very low porosities
- microstructures with features < 2µm

Deposition Efficiency (DE), Mech. Properties

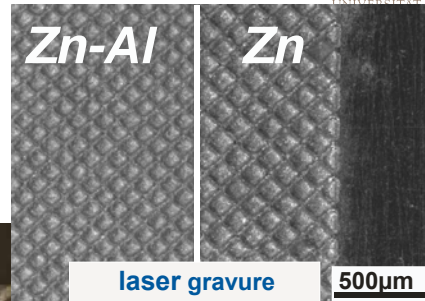
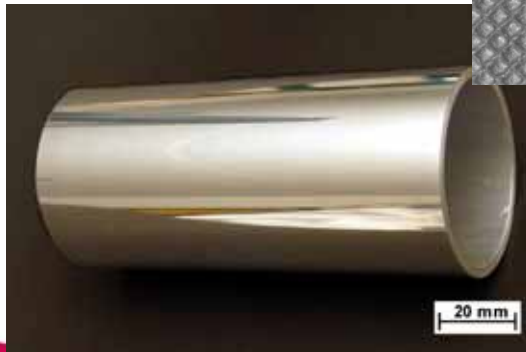


- Zn – Mg6.3 coating is 7 times harder than pure Zn coating
- DE is decreasing with increasing hardness

Prototype Rolls for Laser Engraving



Cold Spray coating
of **Zn** and **Zn - 12Al**



laser gravure

500µm

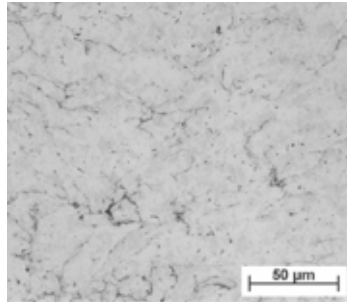
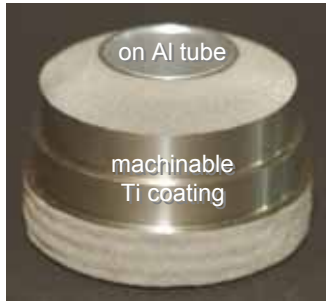
dimples:

- 20 ~ 32µm depth,
- 30 µm width



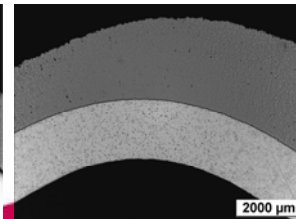
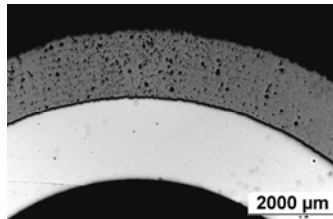
Cold Spray for Rapid Manufacturing

Complex Ti-Parts for Aviation Industry



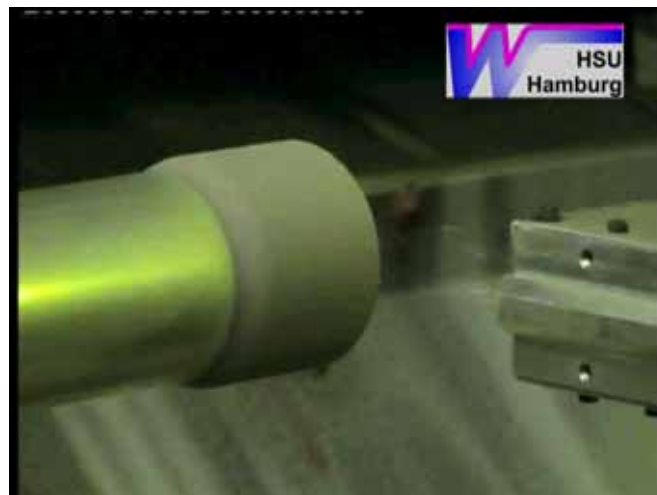
challenge:
impact angle
at contours

goals achieved:
yield strength
465 MPa (tensile test)
< 0.1% porosity
> 95% DE
< 1500 ppm Oxygen

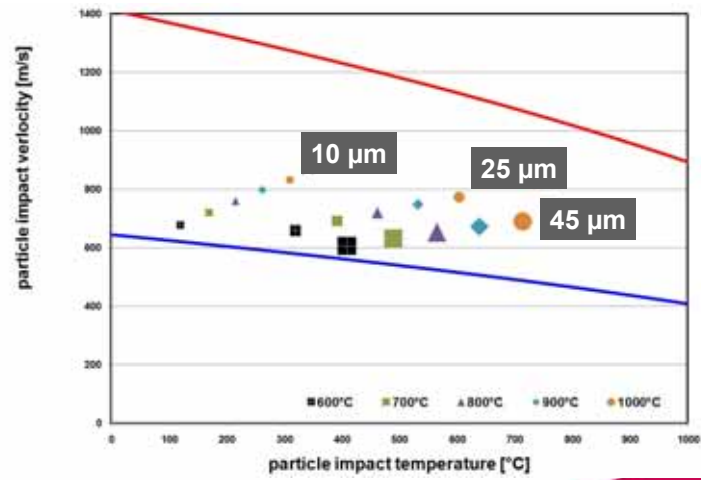


BMBF LuFo-Project together with GE, Linde, CGT

Example: Cold Spraying of Titanium



Impact Conditions for Deposition: Ti

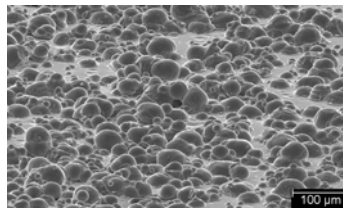


→ exceeding v_{crit} by > 200 m/s

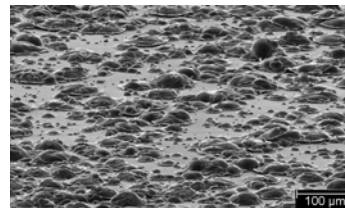
Ti - Coatings: Cold Spray Conditions

$T_{\text{gas}} = 600^{\circ}\text{C}$,
 $p_{\text{gas}} = 4$ MPa

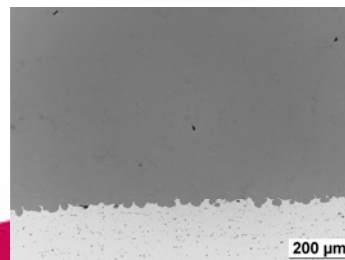
impact
morphologies



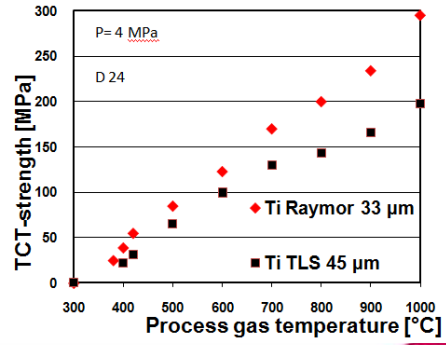
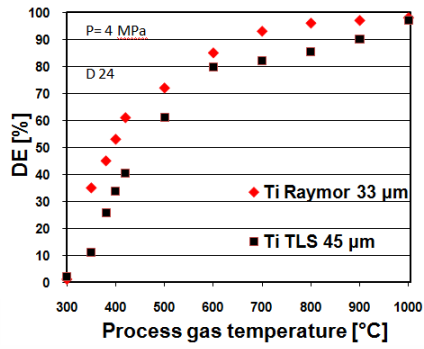
$T_{\text{gas}} = 1000^{\circ}\text{C}$,
 $p_{\text{gas}} = 4$ MPa



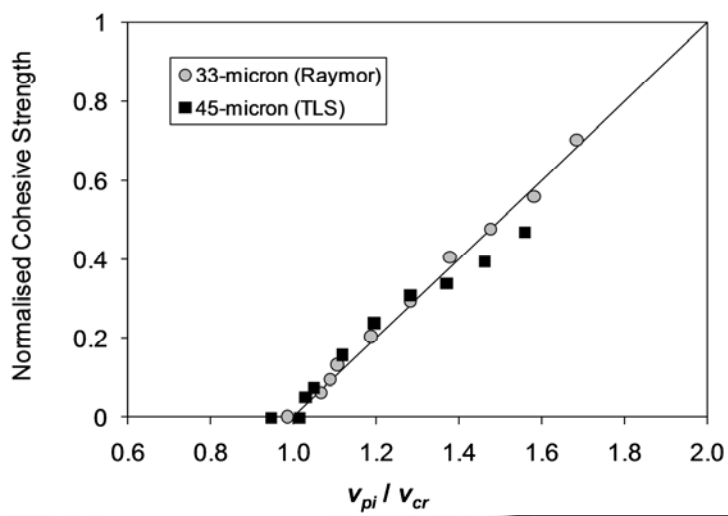
coating microstructure



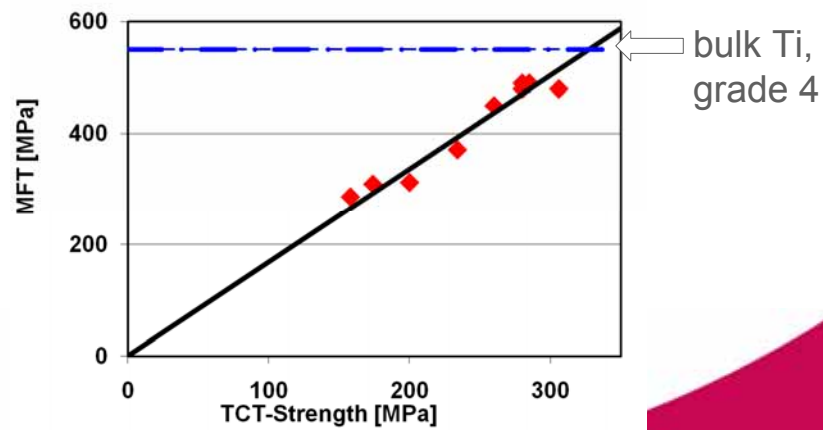
Ti - Coatings: Cold Spray Conditions



Ti: Excess Velocity



Ti: Coating Strengths



Application to Amorphous Alloys ??

heat creation modes by kinetic energy:

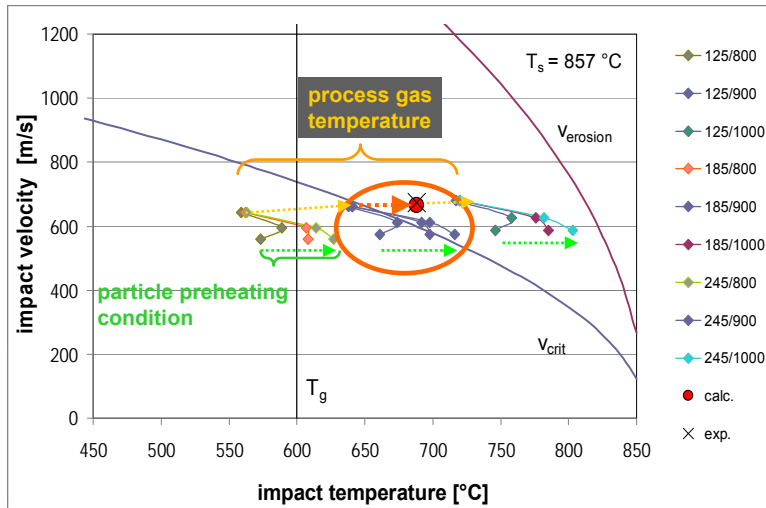
$T < T_g$:
localized shear band formation
by high strain rate deformation

$T > T_g$:
viscous flow

→ localized temperature rise at regions of highest strain rates

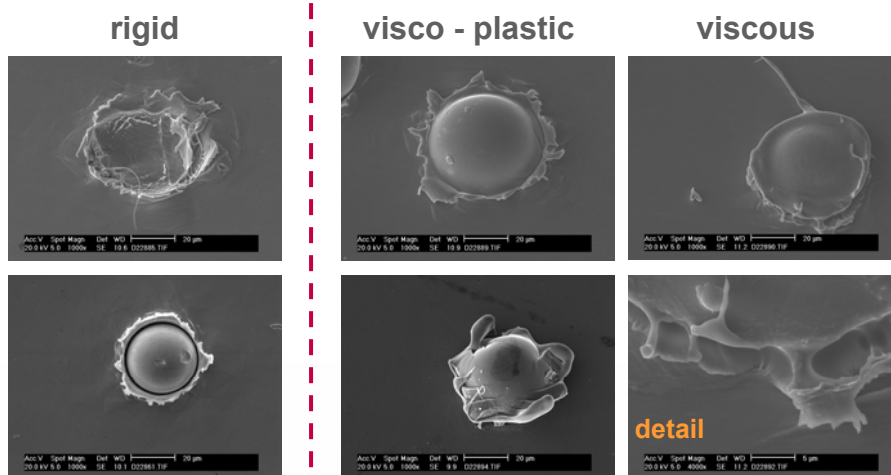
am. $\text{Fe}_{44}\text{Co}_6\text{Cr}_{15}\text{Mo}_{14}\text{B}_{15}\text{C}_6$: $T_s = 860^\circ\text{C} \sim 1.3 T_g$

Impact Conditions, Window of Sprayability: am. FeCoCrMoBC, Size Cut 25- 45 μm



process gas : N_2

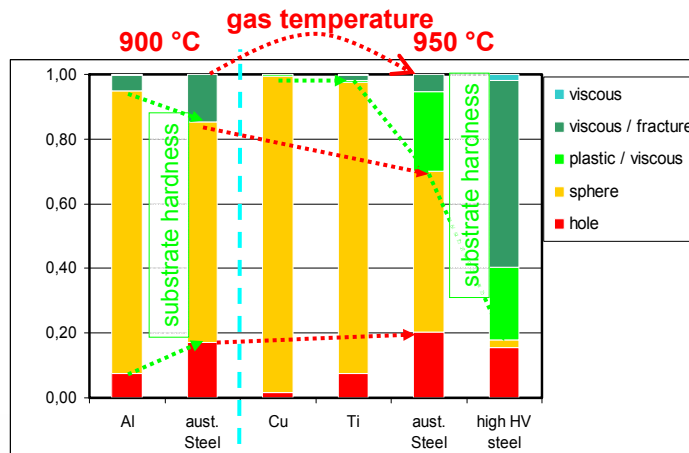
Amorphous $\text{Fe}_{44}\text{Co}_6\text{Cr}_{15}\text{Mo}_{14}\text{B}_{15}\text{C}_6$: Impact Scenarios (on steel 1.4301)



erosive wear

coating formation

Influence of Gas Temp., Substrate Hardness



DE:

26 %

76 %

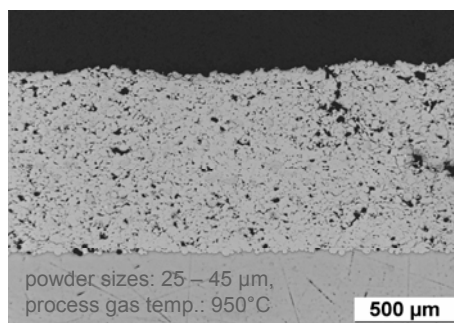
process gas : N₂

more viscous particle deformation by:

higher impact temperature

harder substrate material (higher deformation rates)

Amorphous Fe₄₄Co₆Cr₁₅Mo₁₄B₁₅C₆: Coating Microstructures (on steel 1.4301)



process gas : N₂

crystalline phases: < 2 % (estimated from XRD)

porosity: 2.7 vol. %

hardness: 1100 HV 0.3 (similar to the casted BMG)

tensile strength: 63 MPa

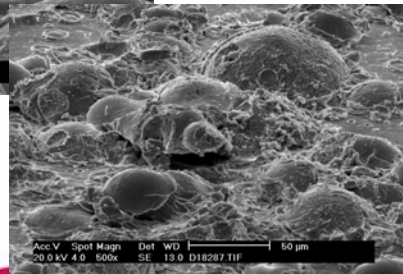
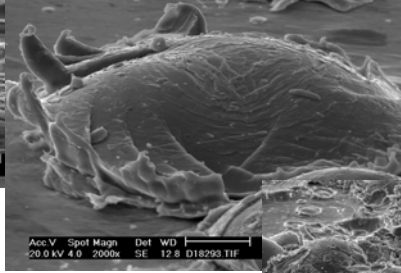
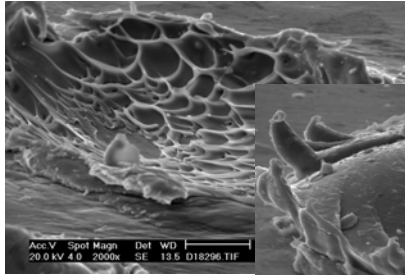
deposition efficiency: 70 %

Impact Scenario of Zr-based BMG on Steel



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powder sizes: $< 25 \mu\text{m}$;
 $T_{\text{impact}}: \sim 50 \text{ }^\circ\text{C (He)}$;
 $V_{\text{impact}}: \sim 1000 \text{ m/s}$

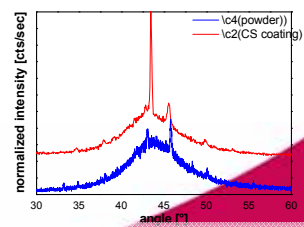
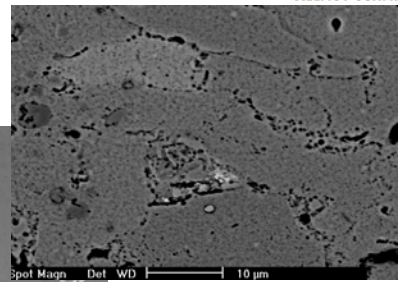
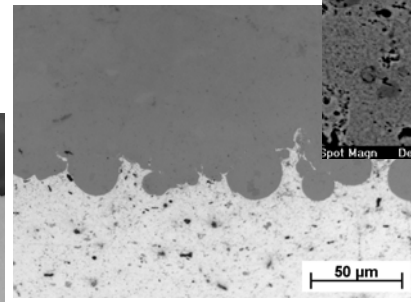
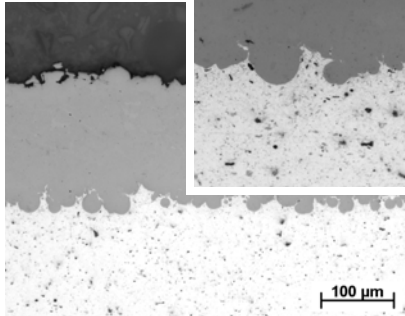


Zr₅₅Al₁₀Ni₅Cu₃₀ (at. %) BMG on AlMg3 (He)



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powder sizes: $< 25 \mu\text{m}$;
 $T_{\text{impact}}: \sim 50 \text{ }^\circ\text{C}$, process gas He;
 $V_{\text{impact}}: \sim 1000 \text{ m/s}$

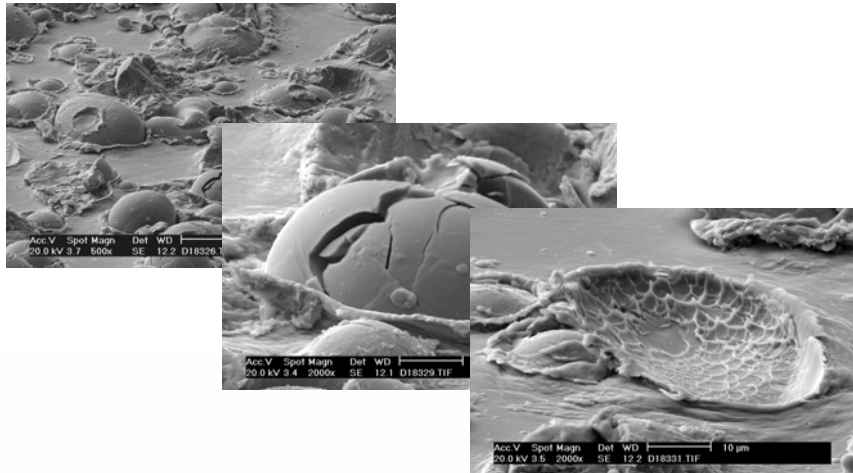


coating:
 amorphous structure: $\sim 90 \%$

porosity: $< 0.2 \%$; hardness: 500 HV 0.3

Limits: Brittleness

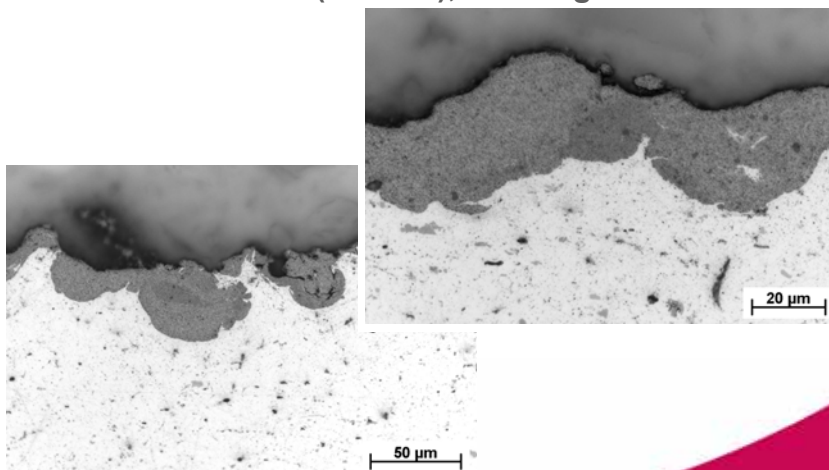
nanocrystalline Zr55 Al10 Ni5 Cu30 (at. %), on steel



→ ductility is needed to create heat and to reach conditions for shear instabilities

Nanocrystalline Alloys on AlMg3

Zr55 Al10 Ni5 Cu30 (in at. %), on AlMg3



Conclusions: Kinetic Spraying



- limited thermal influence:
kinetic spraying is suitable for processing
 - oxidation sensitive materials
 - metastable phases
- bonding by shear instabilities under
ultra high strain rate deformation:
 - applicable to amorphous alloys
- spray conditions depend on materials properties,
particle size and temperature

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