

Olefin-Metathese: Ruthenium-katalysierte C=C -Bindungsknüpfungen



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A decade of Homogeneous Catalysis



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The Nobel Prize in Chemistry 2001

- Nobel Prize Award Ceremony
- William S. Knowles
- Ryoji Noyori
- K. Barry Sharpless



William S. Knowles Ryoji Noyori K. Barry Sharpless

The Nobel Prize in Chemistry 2001 was divided, one half jointly to William S. Knowles and Ryoji Noyori "for their work on *chirally catalysed hydrogenation reactions*" and the other half to K. Barry Sharpless "for his work on *chirally catalysed oxidation reactions*".

The Nobel Prize in Chemistry 2010

- Richard F. Heck
- Ei-ichi Negishi
- Akira Suzuki



Photo: University of Delaware, USA Photo: Purdue University, USA Photo: Hokkaido University, Japan
Richard F. Heck Ei-ichi Negishi Akira Suzuki

The Nobel Prize in Chemistry 2010 was awarded jointly to Richard F. Heck, Ei-ichi Negishi and Akira Suzuki "for *palladium-catalyzed cross couplings in organic synthesis*".

The Nobel Prize in Chemistry 2005

- Nobel Prize Award Ceremony
- Yves Chauvin
- Robert H. Grubbs
- Richard R. Schrock



Photo: U. Montan

Yves Chauvin



Photo: R. Paz

Robert H. Grubbs

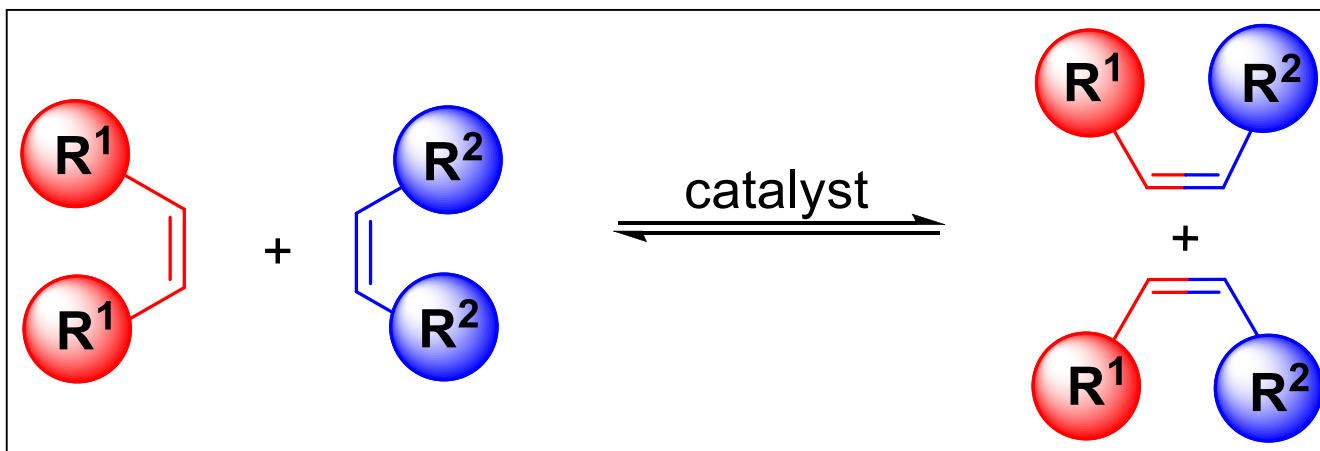
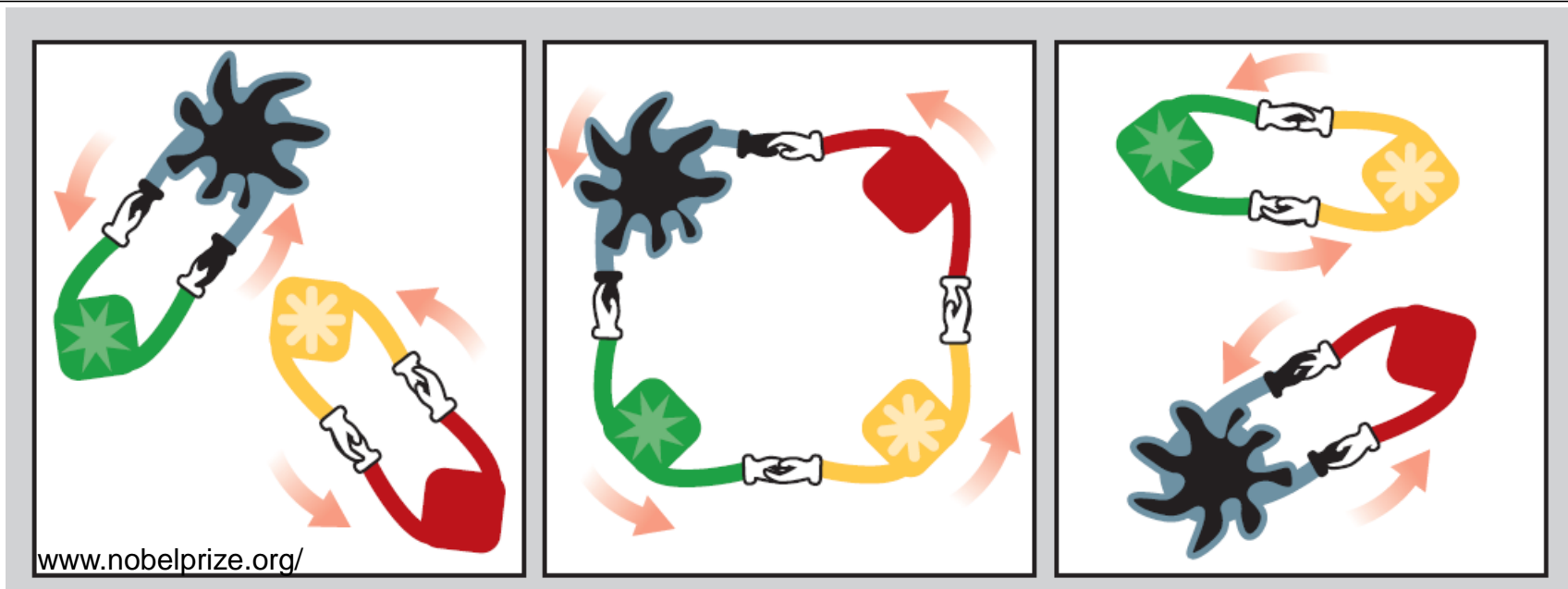


Photo: L.B. Hetherington

Richard R. Schrock

The Nobel Prize in Chemistry 2005 was awarded jointly to Yves Chauvin, Robert H. Grubbs and Richard R. Schrock "for the development of the metathesis method in organic synthesis".

Olefin Metathesis (changing places)



H. S. Eleuterio - Du Pont 1957



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BUNDESREPUBLIK DEUTSCHLAND

KL. 39 c 25/01

DEUTSCHES  PATENTAMT

INTERNAT. KL. C 08 f

AUSLEGESCHRIFT 1 072 811

P 20792 IVb/39 c

ANMELDETAG: 3. JUNI 1958

BEKANNTMACHUNG
DER ANMELDUNG
UND AUSGABE DER
AUSLEGESCHRIFT:

7. JANUAR 1960



ren zur Poly-

1 Metallen der
ems auf inerten
1 Behandeln mit
5 ktive Katalysa-
lekularen festen
n. Die Aktivität
itz eines Hydri-
10 III des Periodi-
rhöht. Die Her-
ter anderem in
47, 2 726 231,
eschrieben.

15 B die Aktivität
nerisation end-
monomerer, wie

20 talsatoren aus
VI a auf einem
g durch Behan-
n Gas, bei Um-
telle des Gas

Verfahren zur Polymerisation cyclischer,
insbesondere mono-,
bi- oder tricyclischer Olefine

Anmelder:

E. I. du Pont de Nemours and Company,
Wilmington, Del. (V. St. A.)

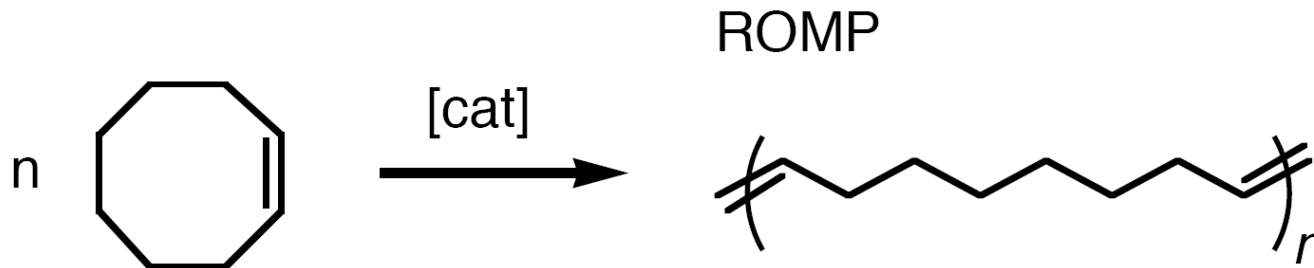
Vertreter: Dr.-Ing. W. Abitz, Patentanwalt,
München 27, Gaußstr. 6

Beanspruchte Priorität:

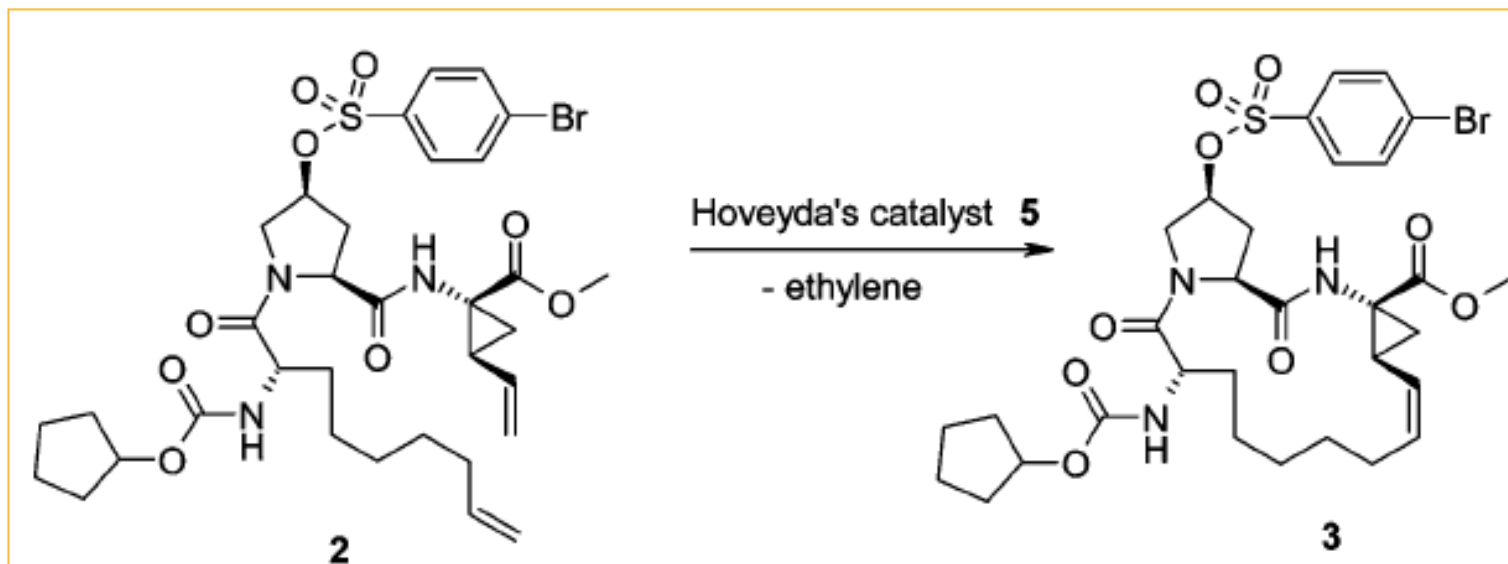
V. St. v. Amerika vom 20. Juni 1957

Herbert Sousa Eleuterio, Wilmington, Del. (V. St. A.),
ist als Erfinder genannt worden

Hüls-Vestenamer-Prozess



HCV Protease Inhibitor – BILN 2061



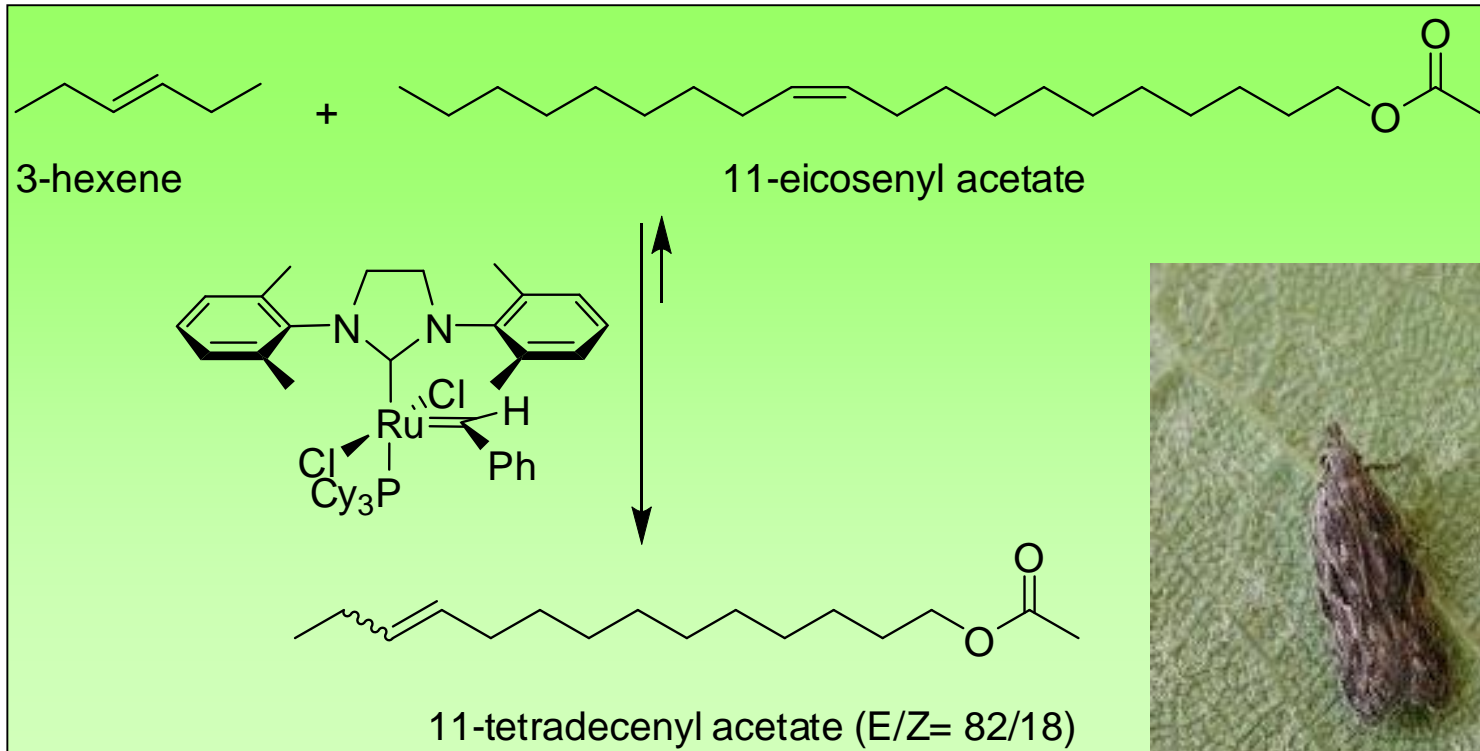
This first-generation RCM was scaled to produce >100 kg of the active pharmaceutical ingredient (API).

.....optimization of the initial RCM, leading to a second-generation process, which dramatically improves throughput and reaction efficiency.

Second-Generation Process for the HCV Protease Inhibitor BILN 2061: A Greener Approach to Ru-Catalyzed Ring-Closing Metathesis, Boehringer Ingelheim, *Org. Process Res. Dev.* **2009**, *13*, 250.

The Growing Impact of Catalysis in the Pharmaceutical Industry, *Adv. Synth. Catal.* **2011**, *353*, 1825; C. H. Senanayake et al. doi: 10.1002/adsc.201100488

Pheromones



omnivorous leafroller

Speciality Chemicals, Olefins, Oleochemicals from Palm Oil

Elevance Renewable Sciences and Wilmar International Limited Begin Commercial Shipment of Specialty Chemicals from New World-Scale Biochemical Refinery in Asia

July 18, 2013 (Woodridge, IL)

Elevance-Wilmar Joint Venture Produces Renewable Performance Specialty Chemicals, Olefins and Oleochemicals

Woodridge, Ill., USA/Singapore (July 18, 2013) – Wilmar International Limited, Asia's leading agribusiness, and Elevance Renewable Sciences, Inc., a high-growth specialty chemicals company, announced today that they have begun shipping commercial products, including novel specialty chemicals, to customers from their first world-scale joint venture biorefinery, located in Gresik, Indonesia. The biorefinery is the first based on Elevance's proprietary metathesis technology.

production sites

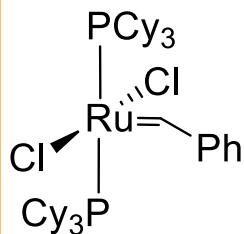
180.000 t/a Gresik, Indonesia 2013
+ Natchez, Mississippi 2016?



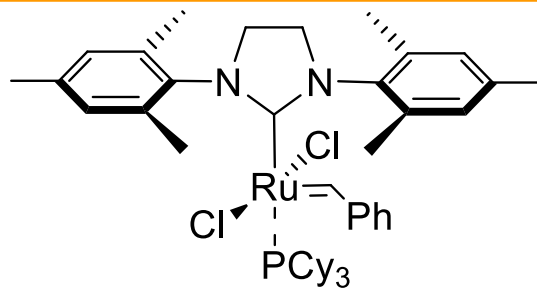
Olefin metathesis precatalysts



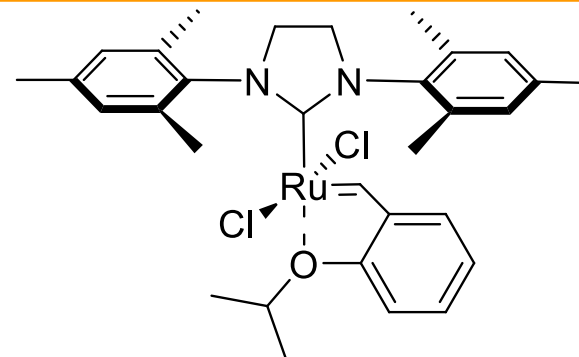
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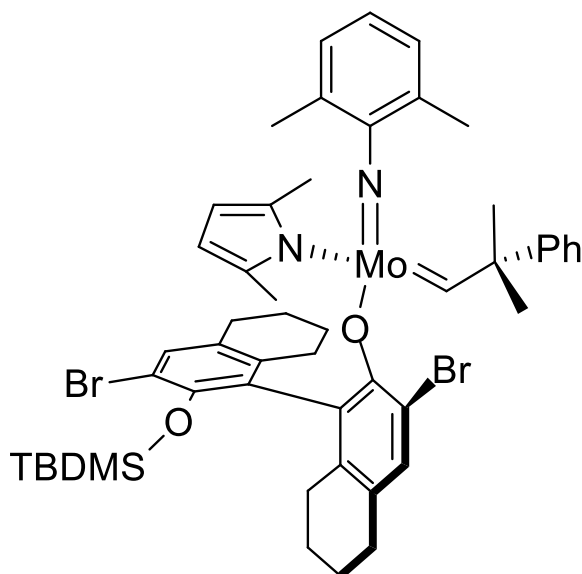
Grubbs I



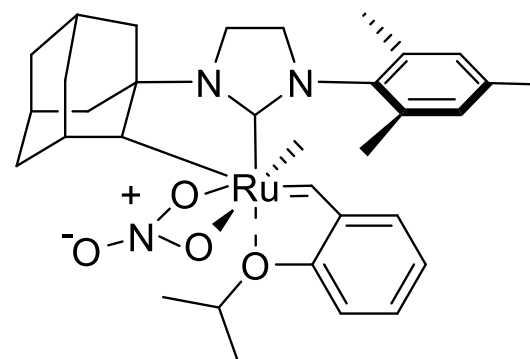
Grubbs II



Grubbs-Hoveyda

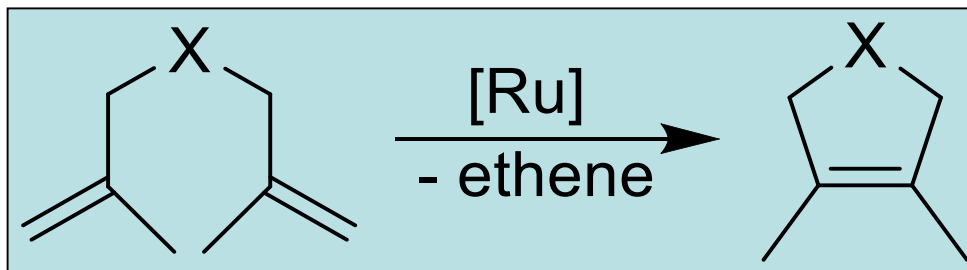


Hoveyda-Schrock



Grubbs

Ring Closing Metathesis leading to sterically-demanding alkenes

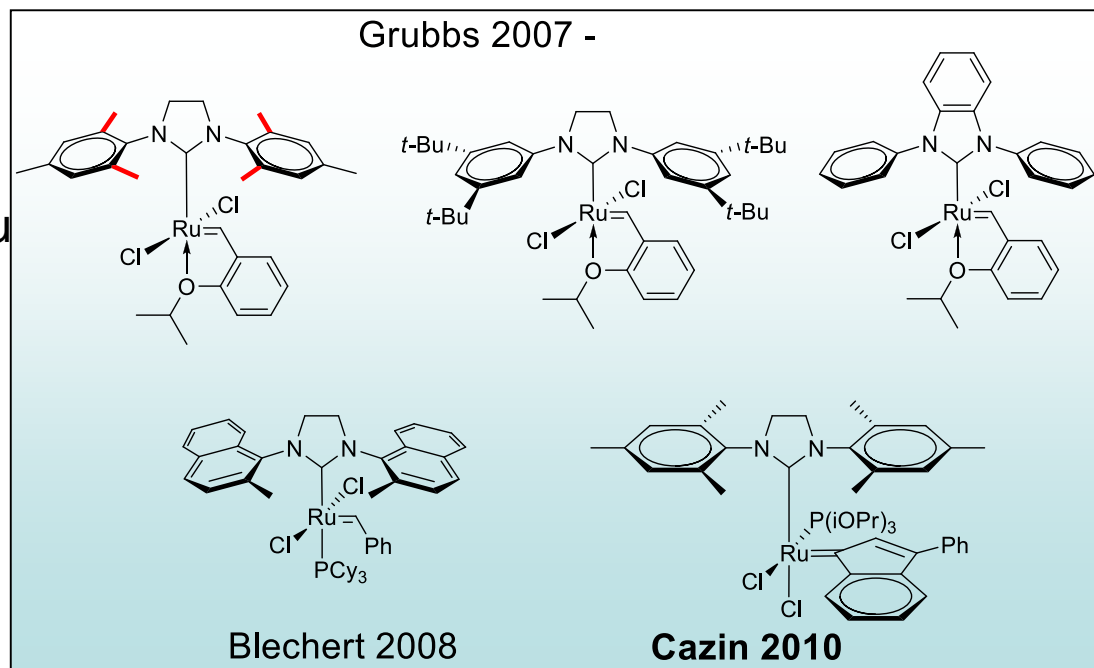


- Problems:**
- excessive catalyst loading (5 - 10 mol%)
 - modest product formation

How to solve the problem?

→ decreased steric bulk around Ru

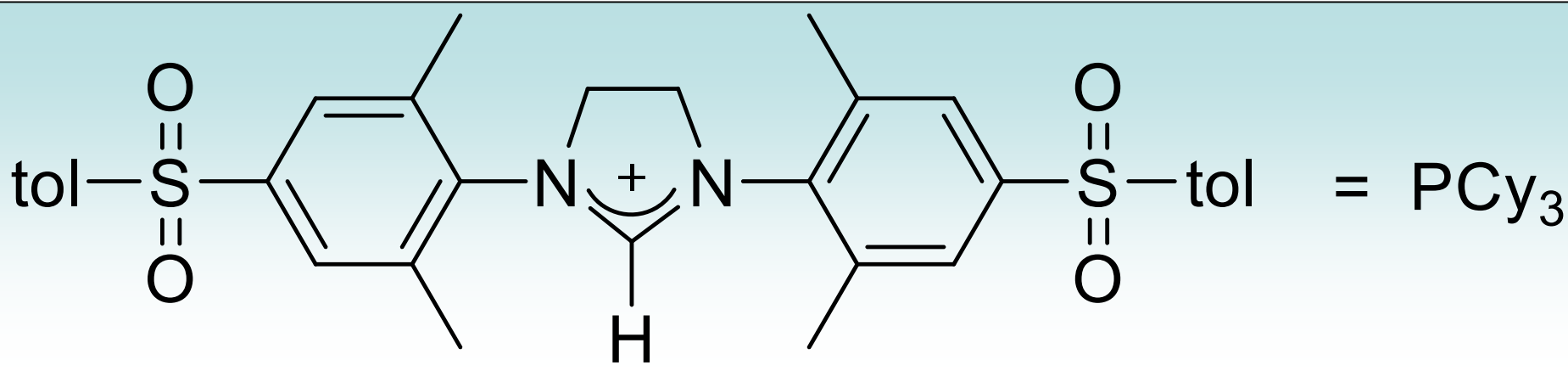
→ Increased precatalyst stability



Weakly donating *N*-heterocyclic Carbenes

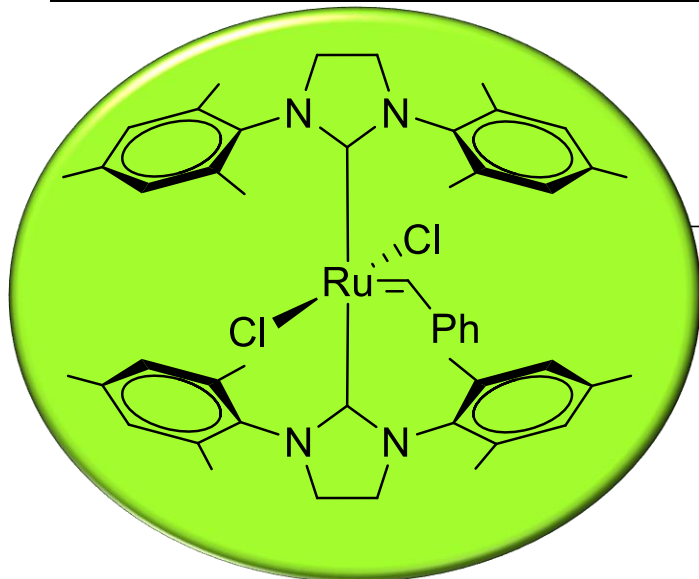


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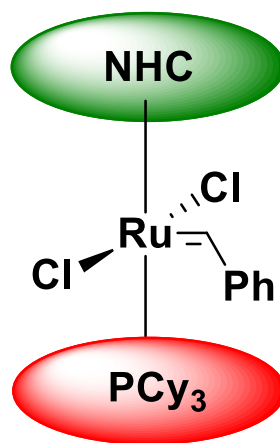
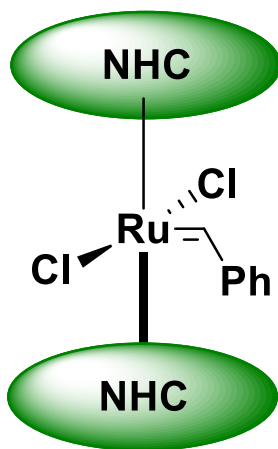


NHC_{ewg}

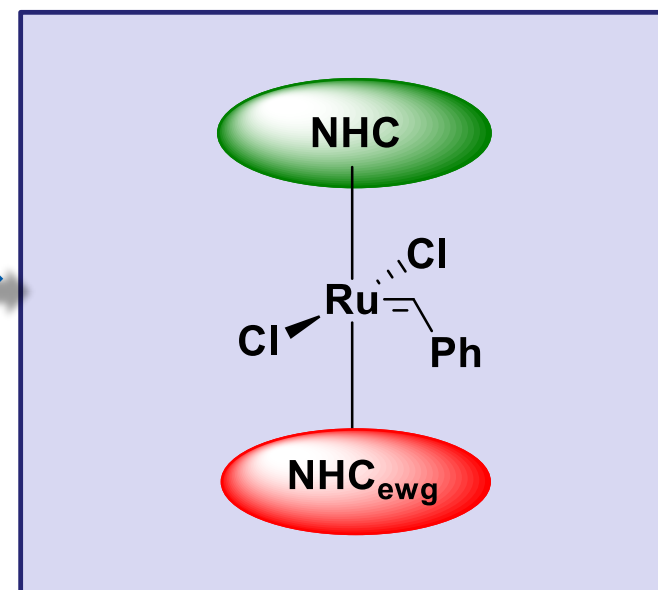
bisNHC complexes



Herrmann, Grubbs 1999
modest activity



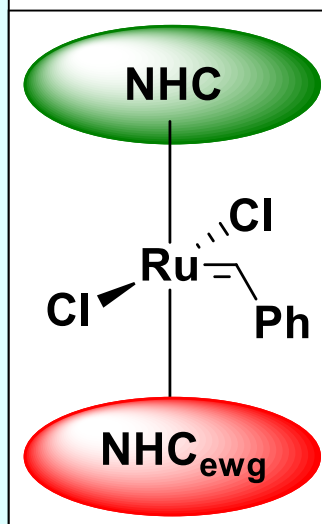
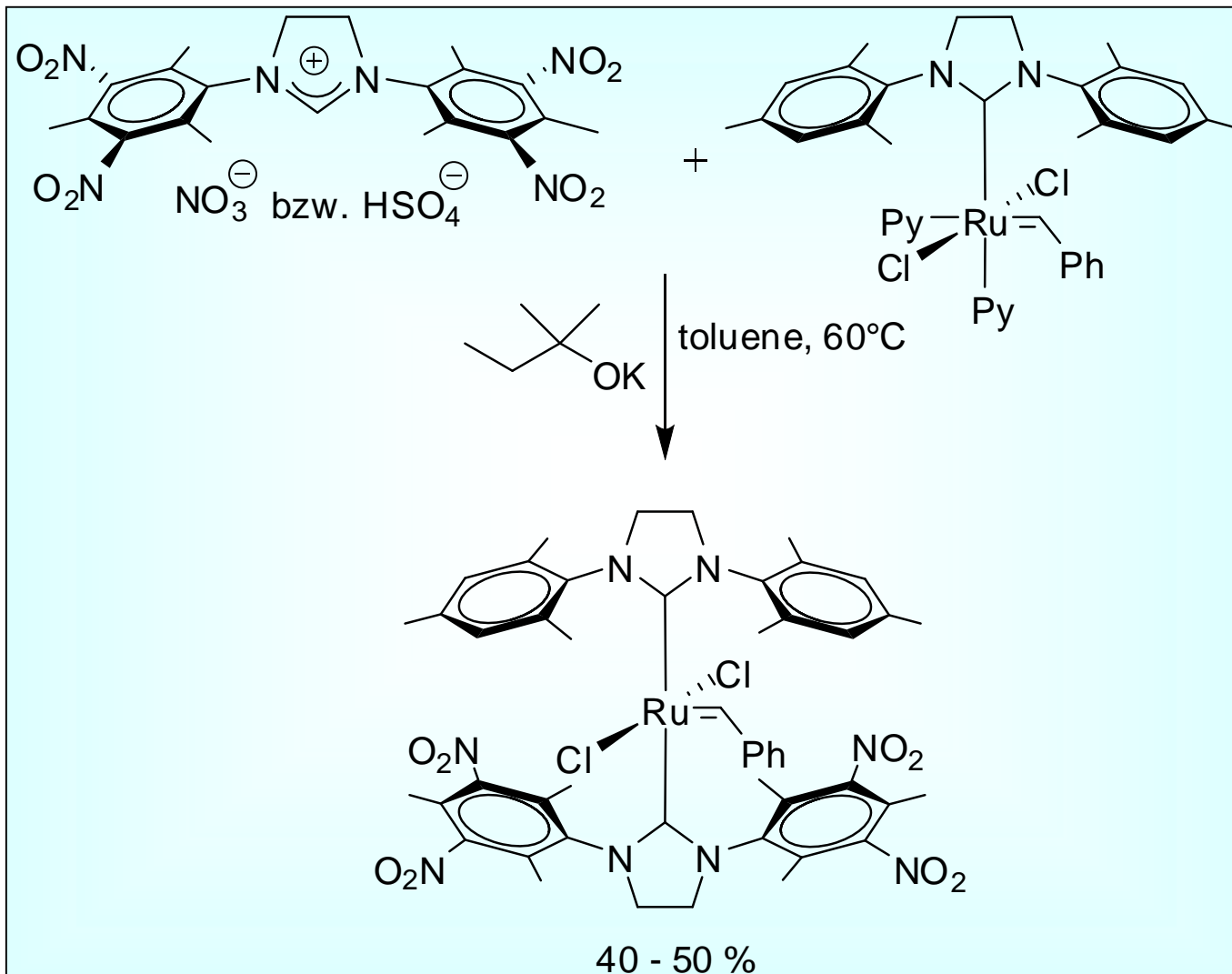
Grubbs II - Komplex



Synthesis of (NHC)(NHC_{ewg})RuCl₂(CHPh)



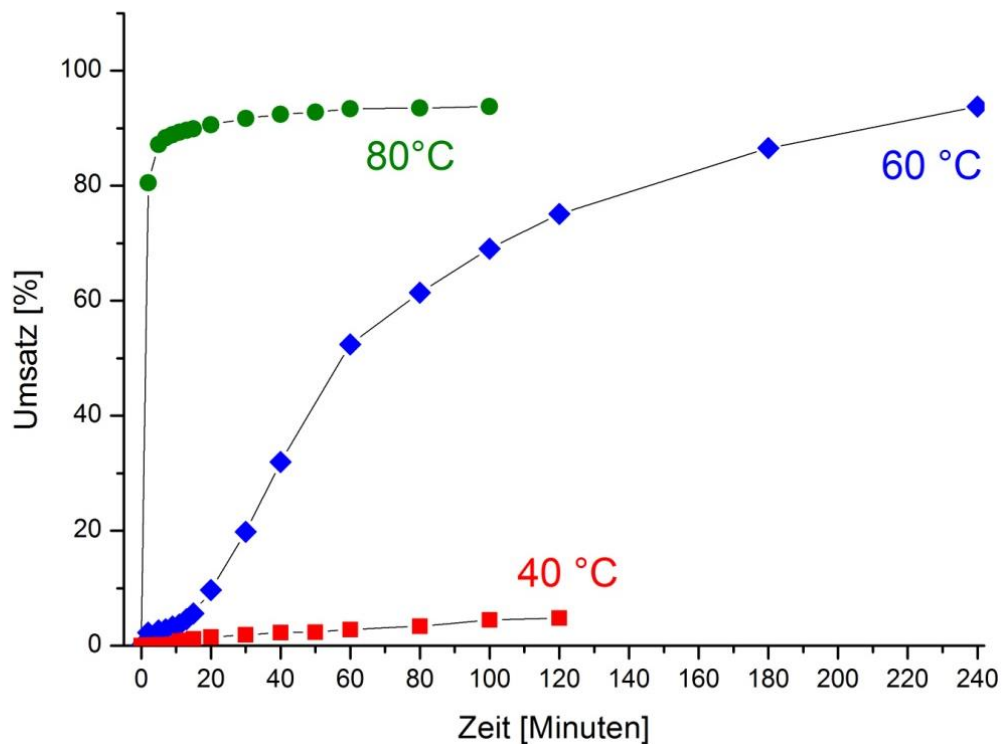
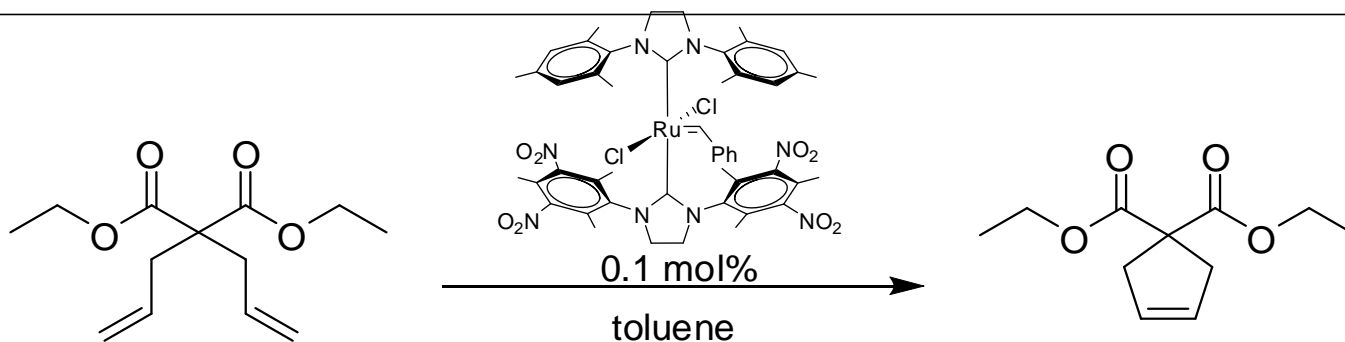
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Test of Ring-closing Metathesis Activity



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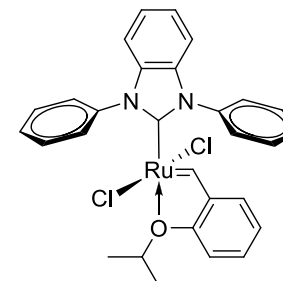
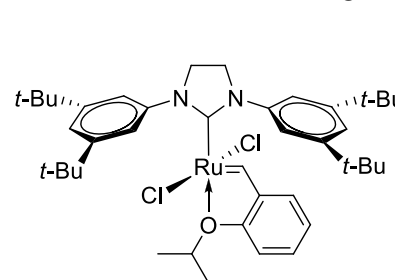
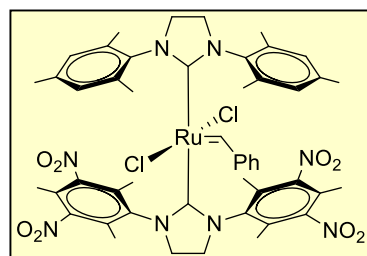


Selected examples for sterically demanding substrates

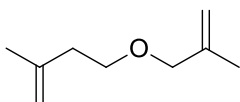
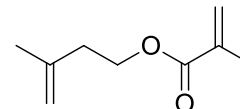
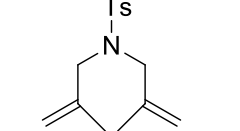
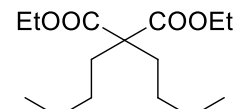

80°C Toluol - 0.5 mol%

60°C C₆D₆ - 5 mol%

Vorfalt 2009



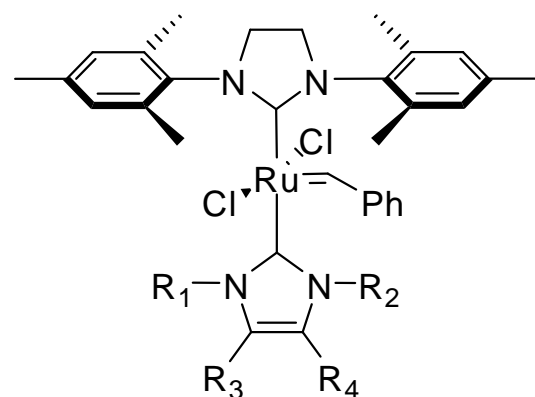
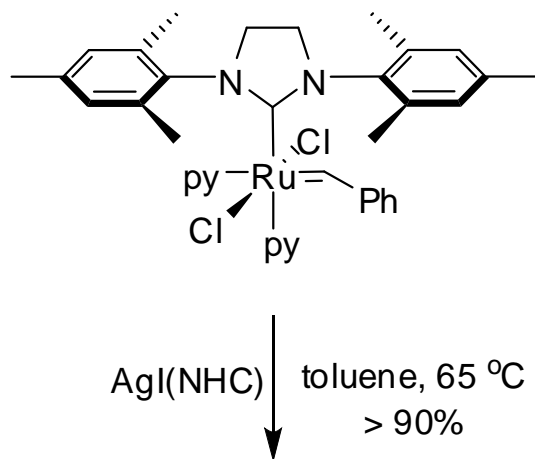
Grubbs 2007

	98%	78%	< 1%
	26%	< 1%	< 1%
	98%	> 95%	> 95%
	60%	51%	36%
	87% (1 mol%)		

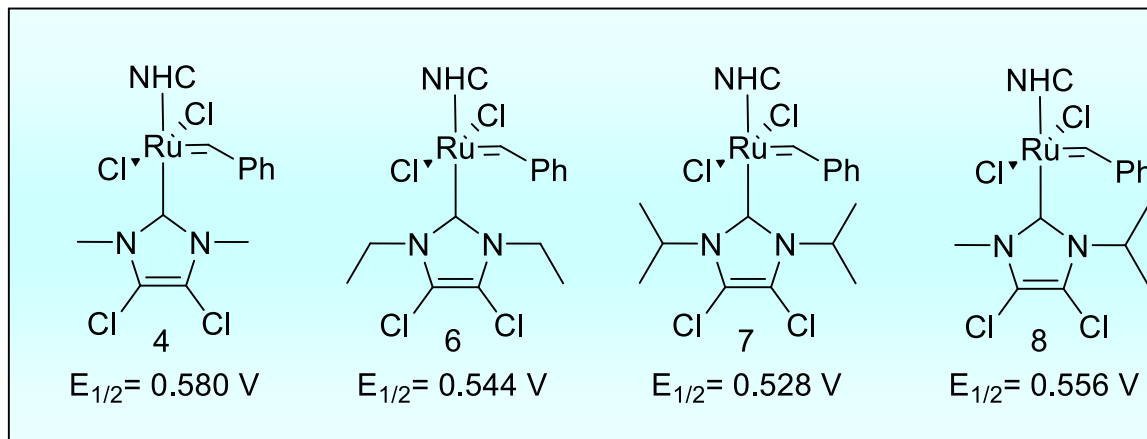
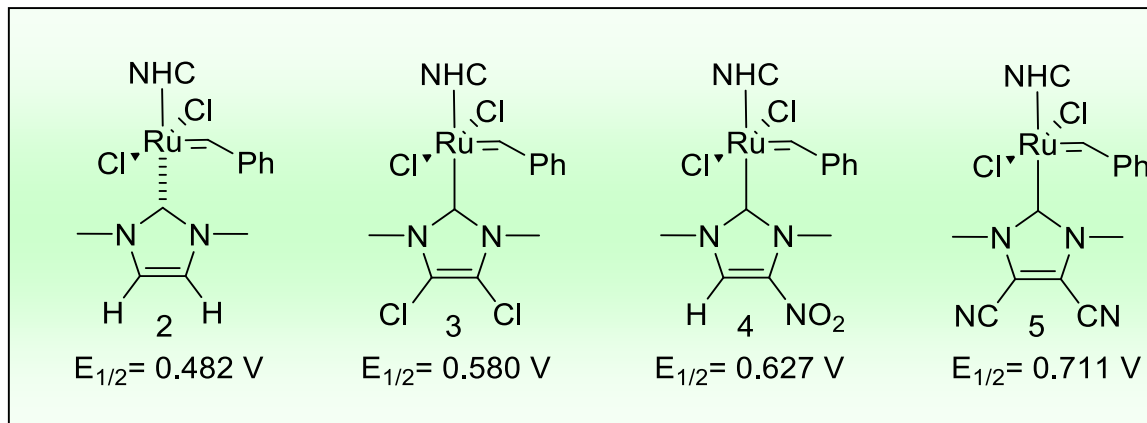
Optimized (NHC)(NHC_{ewg})RuCl₂(CHPh)



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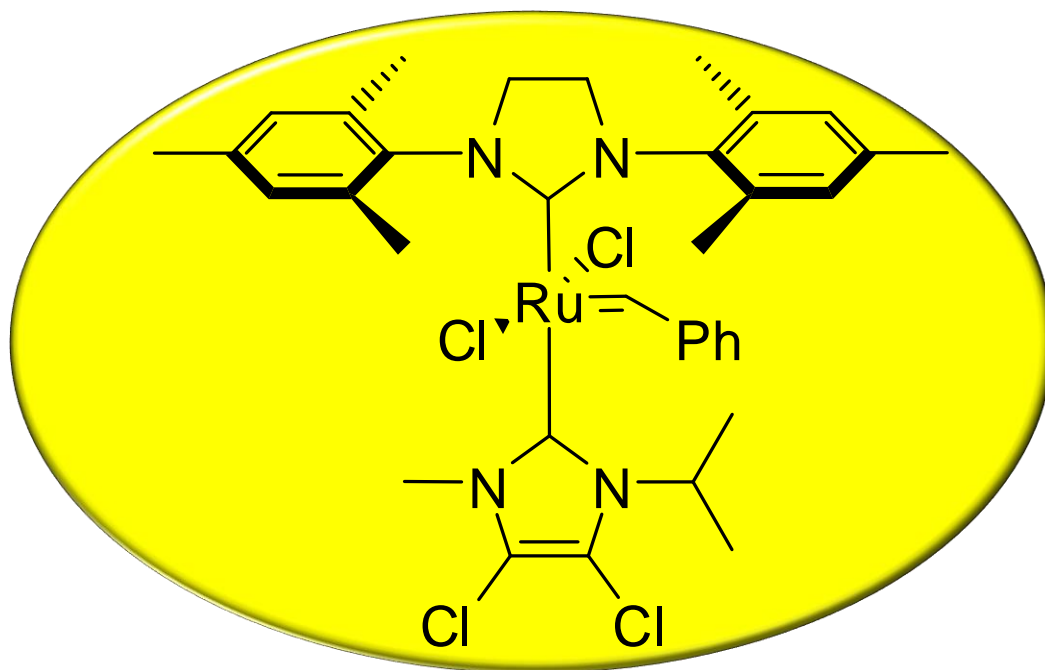
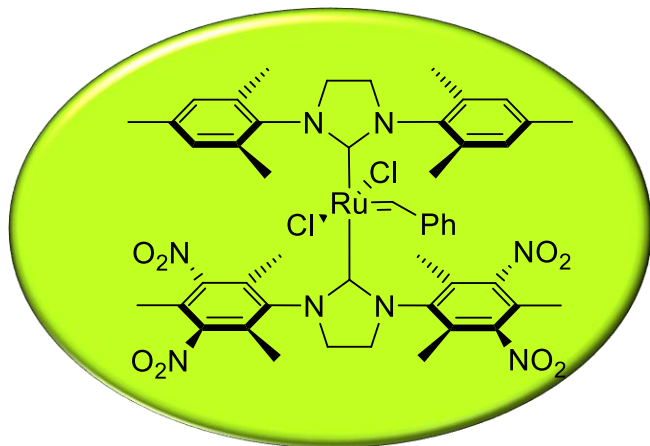


electronic series



steric series

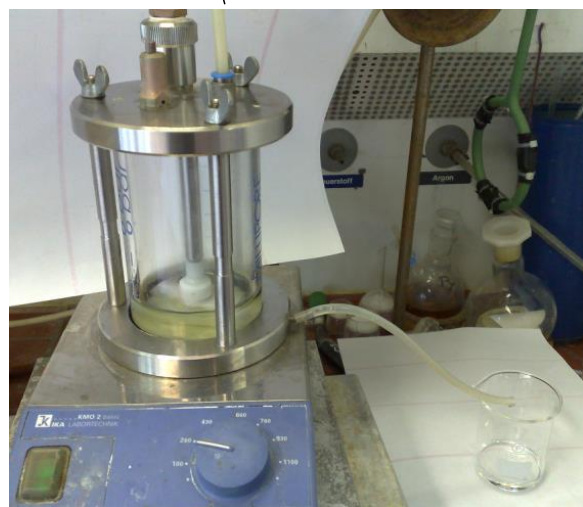
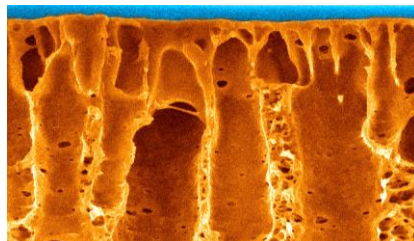
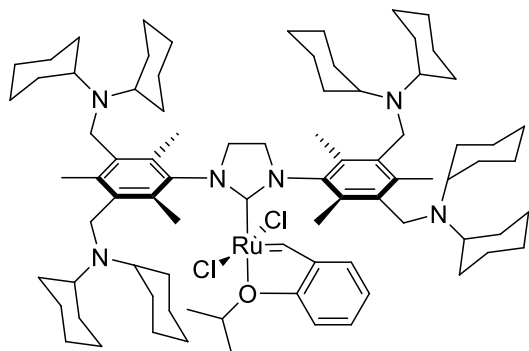
catalytic activity ...ca. three times higher activity.....



V. Sashuk, L. H. Peeck, H. Plenio, *Chem. Eur. J.* **2010**, *16*, 3983-3993
L. H. Peeck, H. Plenio, *Organometallics* **2010**, *29*, 2761-2768
S. Wolf, H. Plenio, *J. Organomet. Chem.* **2010**, *695*, 2418-2422.

Catalyst Separation

Solvent resistant Nanofiltration



$\Delta p = 5 \text{ bar}$

1.4 mL/min



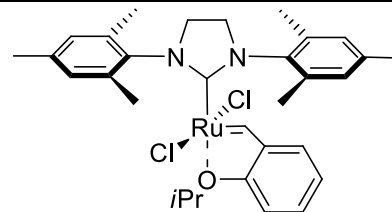
- 99.9% retention of catalyst
- < 4 ppm Ru content

Batchwise and continuous organophilic Nanofiltration of Grubbs type Olefin Metathesis Catalysts.

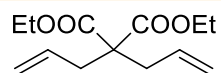
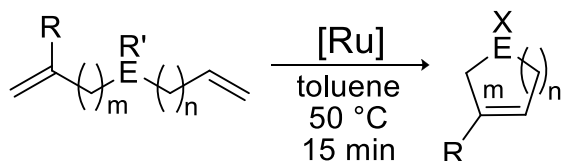
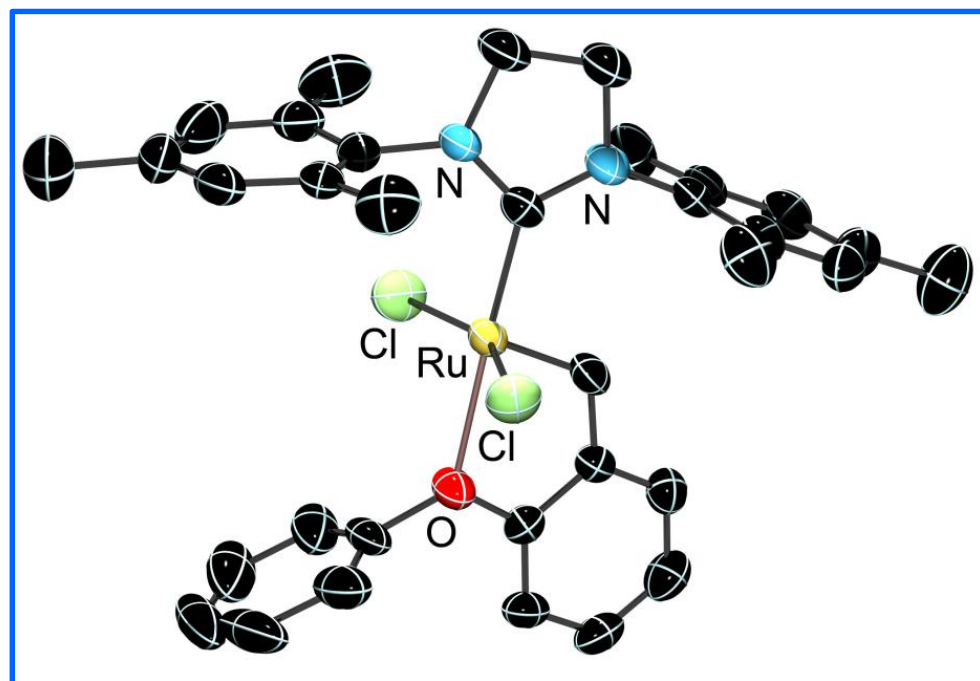
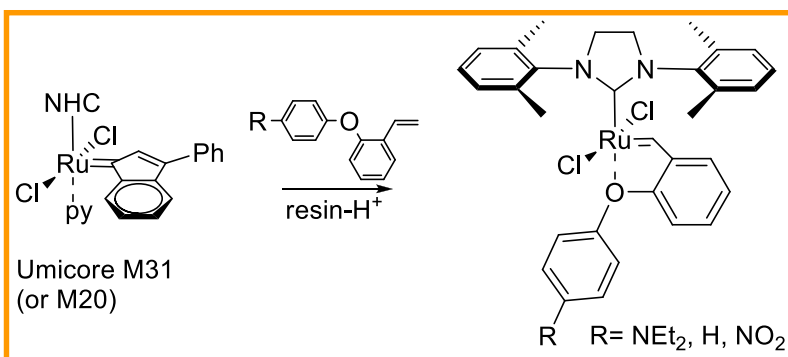
D. Schoeps, K. Buhr, K. Ebert, H. Plenio, *Chem. Eur. J.* **2009**, *15*, 2960.

A. Datta, K. Ebert, H. Plenio, *Organometallics* **2003**, *22*, 4685 for Palladium.

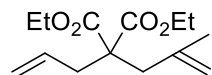
fast RCM catalysis



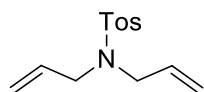
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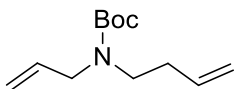
100 ppm 88%



100 ppm 89%



15 ppm 91%

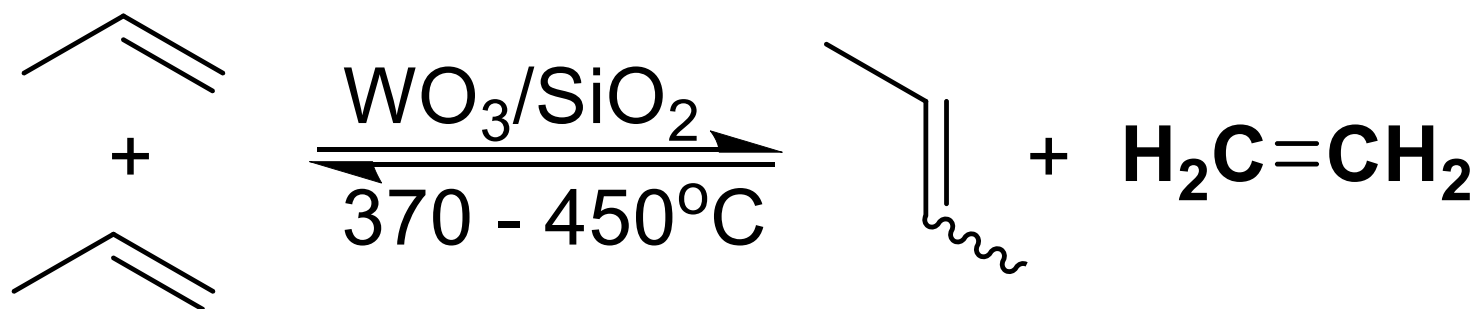


50 ppm 89%

Industrial Olefin Cross Metathesis



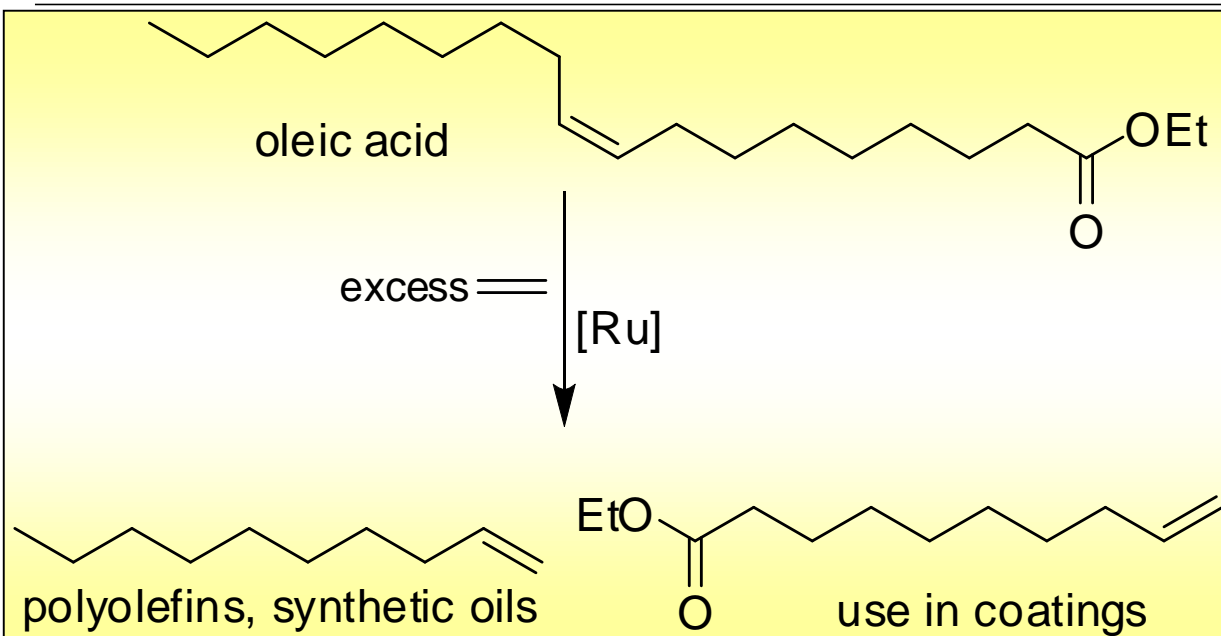
Phillips-Triolefin Process, -1972



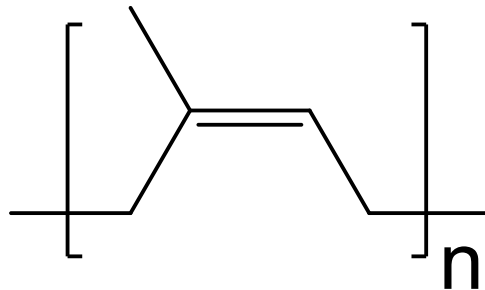
Olefin-Conversion Technology, BASF-FINA



Plant oils for ethenolysis



Natural Rubber



cis-Polyisoprene
Hevea brasiliensis

Annual production ca. $10 \cdot 10^6$ t/a
Philippines, Thailand, Malaysia
1.5 – 3 US\$ /kg (2009/2010)

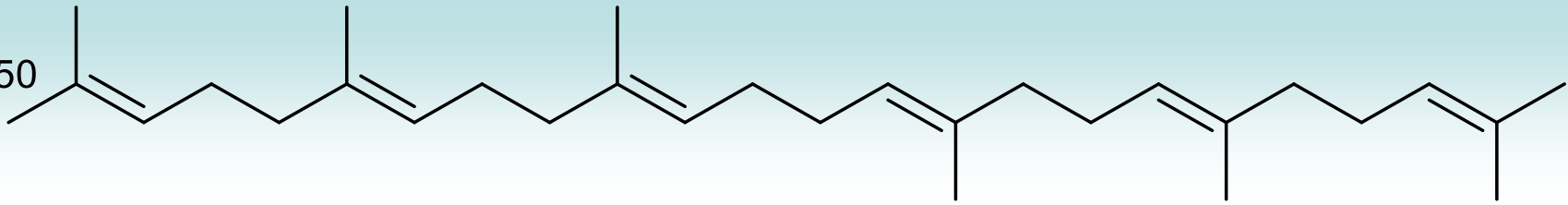


Squalene – a simple model for Natural Rubber

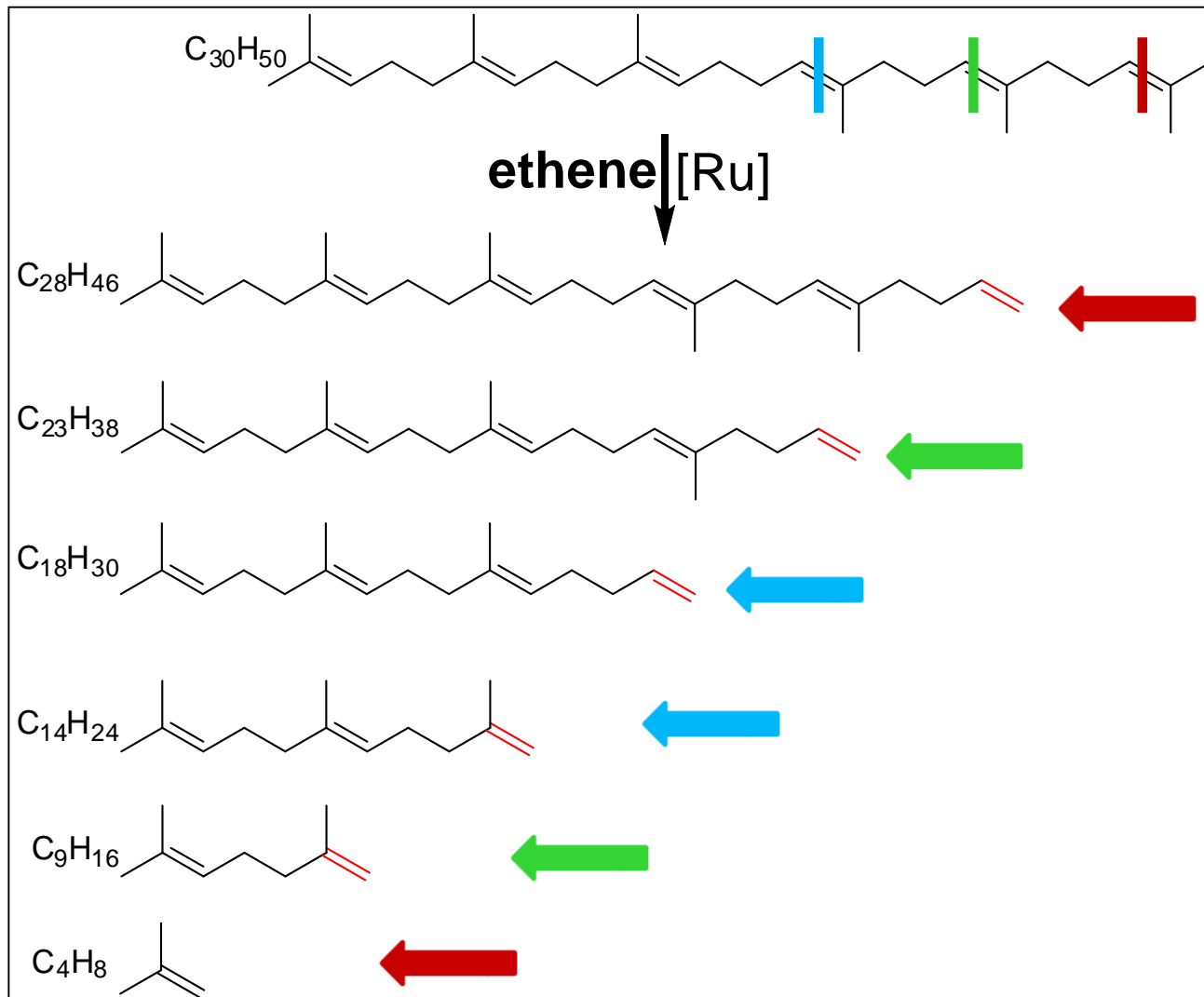


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$C_{30}H_{50}$



Squalene Ethenolysis (cross metathesis with ethene)



Ethenolysis of natural rubber

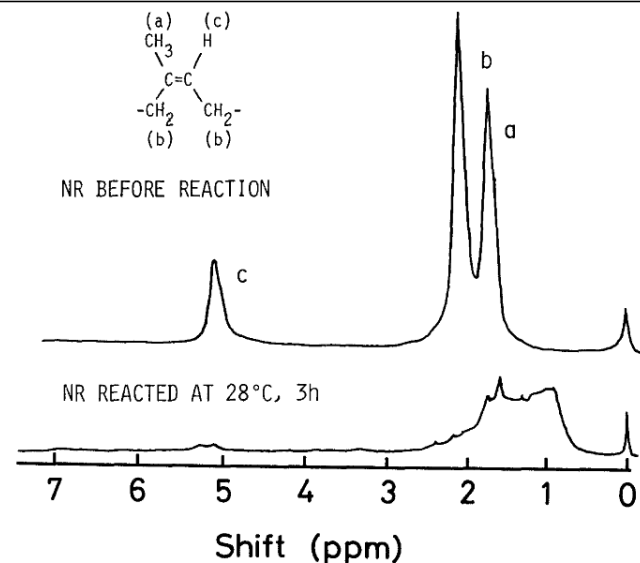


Ethenolysis of Natural Rubber and Polyisoprene

malaysian natural rubber

Catalyst: $WCl_6/SnMe_4$

A. Alimuniar, M. A. Yarmo, M. Z. A. Rahman, S. Kohjiya, Y. Ikeda, S. Yamashita, *Polymer Bull.*, **1990**, 23, 119



“A significant decrease in the polymer’s molecular weight is observed”

Poly-trans-isoprene

Catalyst $[(CF_3)_2CH_3CO)(NAr)W=CHtBu]$

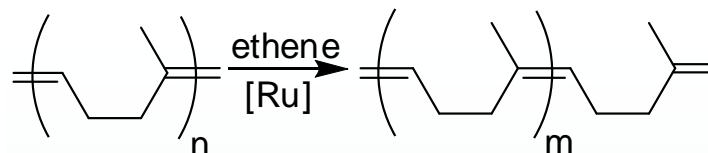
2.33 mol% per double bond

Tab. 1. Depolymerization results

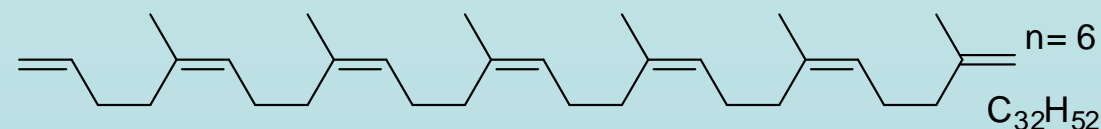
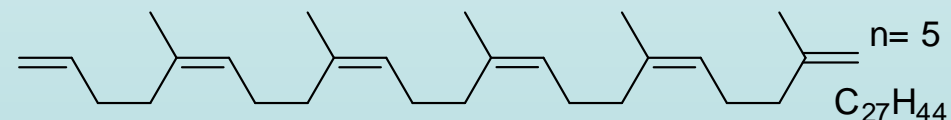
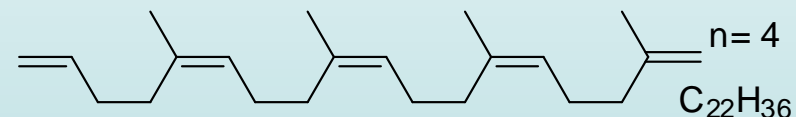
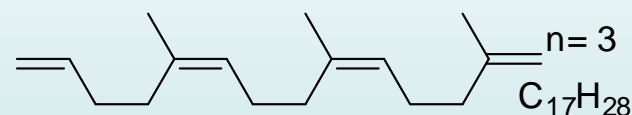
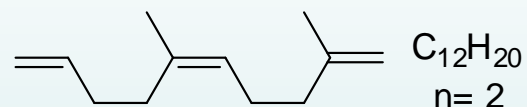
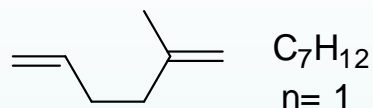
Polymer	\bar{M}_w	\bar{M}_w/\bar{M}_n	Method	MW of products	Detected monomer
Polyoctenamer	108 000 ^{a)}	(1,9)	A	1 000 ^{c)}	1,9-Decadiene ^{f)}
Polybutadiene	200 000 ^{b)}	(2,1)	B, C	2 000 ^{d)}	1,5-Hexadiene ^{f, g)}
	200 000 ^{b)}	(2,1)	C(THF)	1 000 ^{e)}	1,5-Hexadiene ^{f, g)}
Polyisoprene	$2,5 \cdot 10^5$ ^{b)}	(1,0)	C	1 000 ^{e)}	2-Methyl- -1,5-hexadiene ^{f, g)}

K. B. Wagener, R. D. Puts, D. K. Smith,
Makromol. Chem., Rapid Commun. **1991**, 12, 419

Depolymerization of Natural Rubber



oligoisoprenes
m = 1, 2, 3, 4, 5, ...



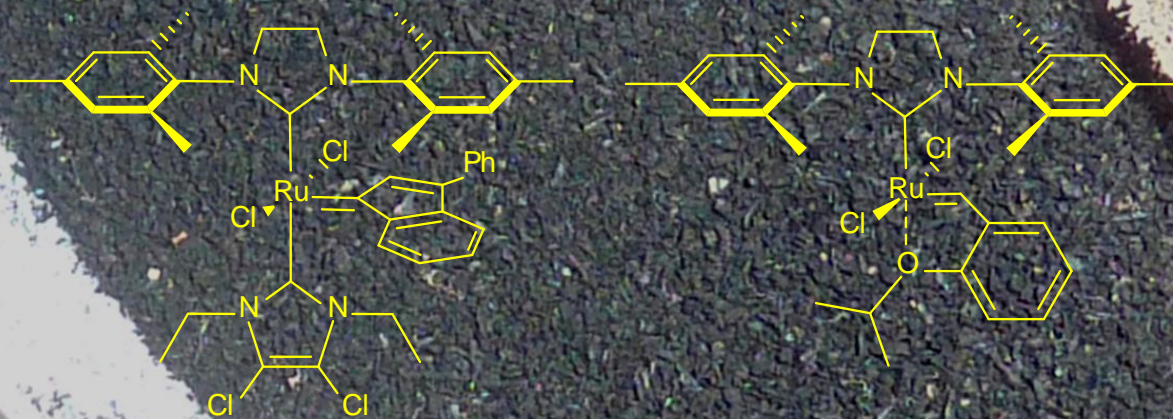
**global annual output: 1.500.000.000 units
end-of-life tires (ELT)**



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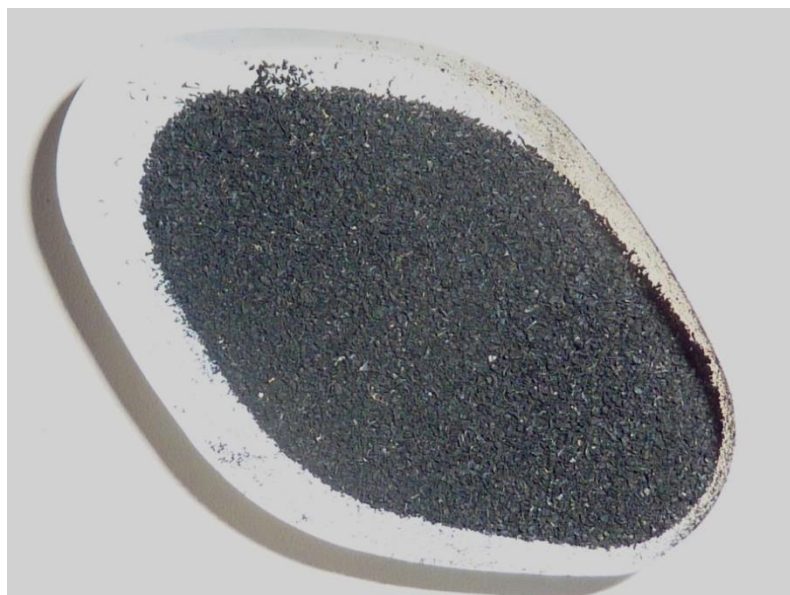


end-of-life tire rubber granulate



end-of-life tire rubber granulate

organic solubles obtained after ethenolysis of ELT powder



[Ru]
80 °C
7 bar
ethene
toluene



10 g of ELT yield up to 5 g of organic solubles

10 g of ELT require 0.04 g of ruthenium complex

Thank You!



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Volodia Sashuk

Lars Peeck

Steffi Wolf

Roman Savka

Pavlo Kos

Tim Vorfalt

Dr. Ebert/GKSS

Deutsche
Forschungsgemeinschaft

DFG



WEBER & SCHAER
DEGR. 1844



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